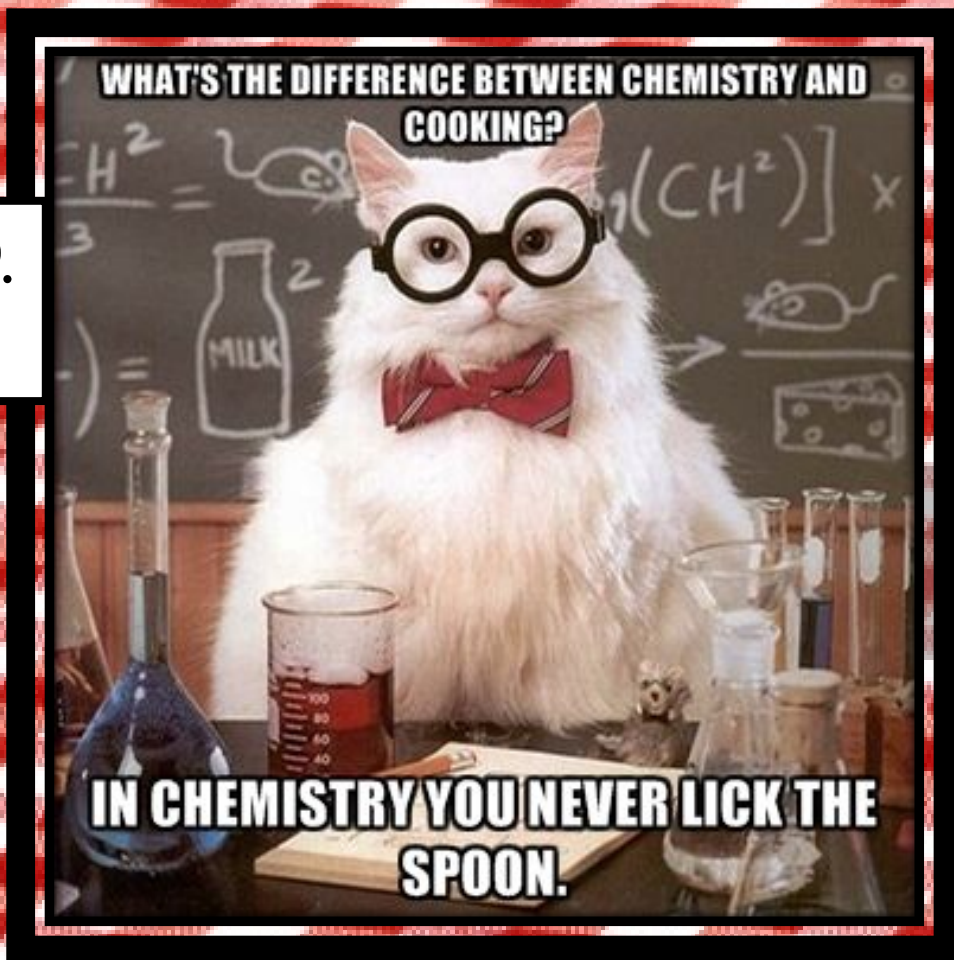


Culinary Chemistry: A Campus Cuisine Cookoff

MSUB Library Lecture series
Spring 16

Michele McMullen R.D.
Dr. Matt Queen





EGGS!

EGGS!

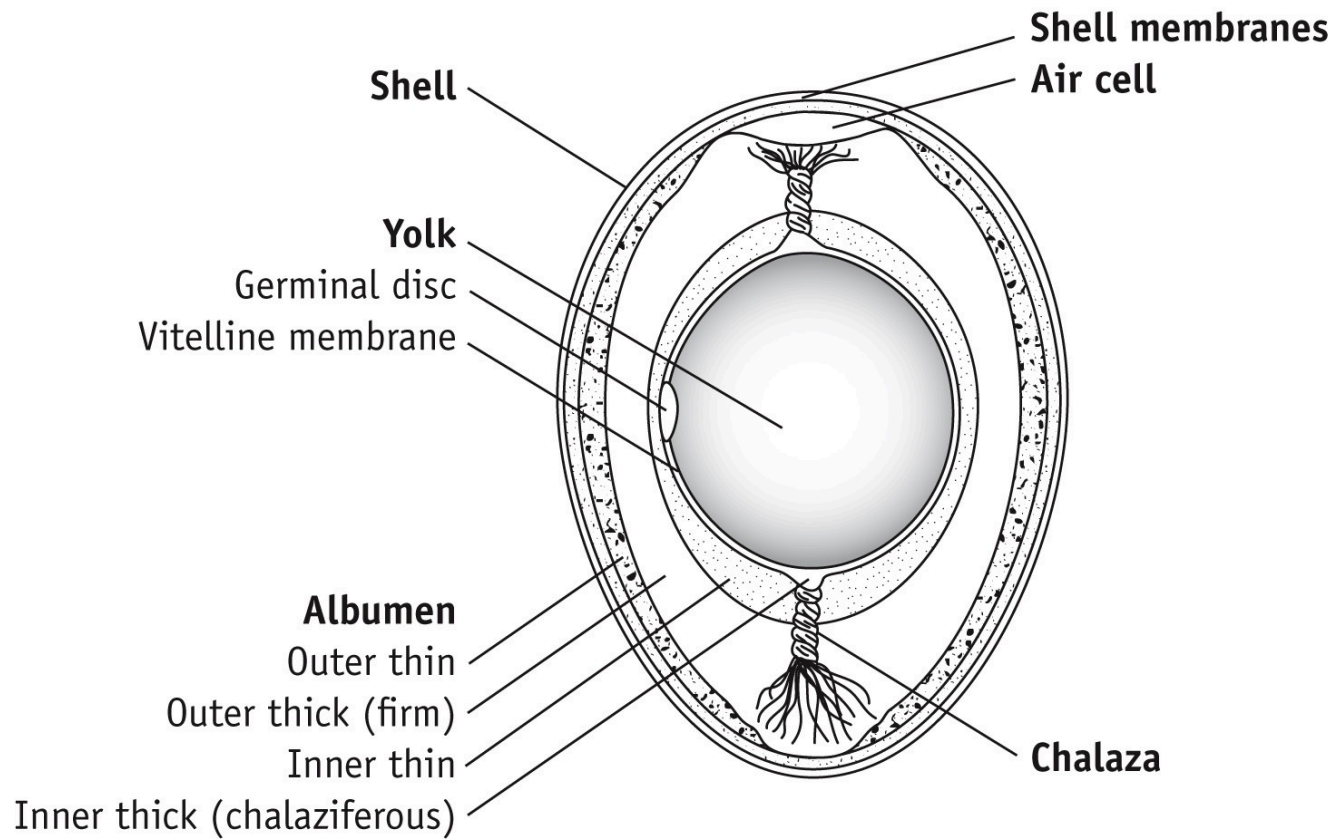
EGGS!

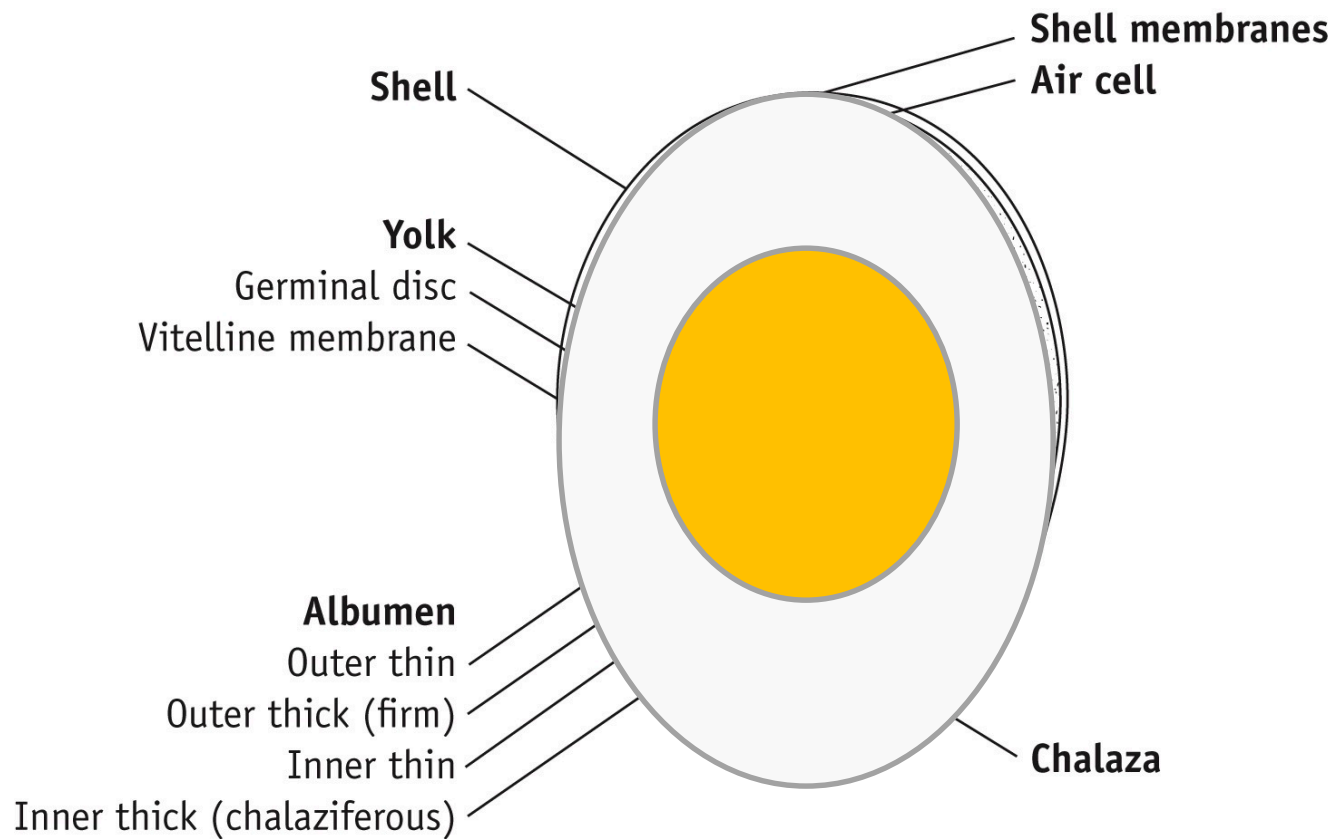
EGGS!

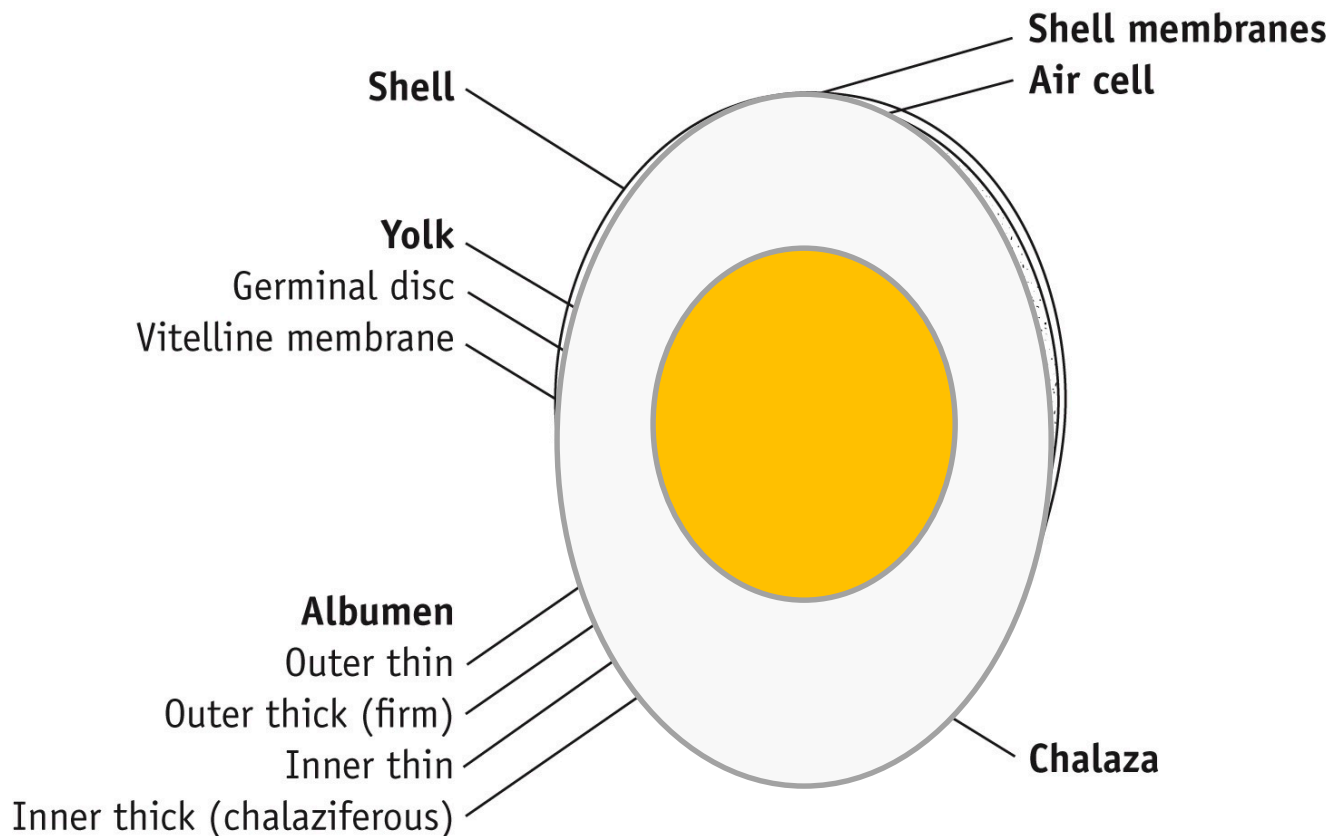
EGGS!

EGGS!

EGGS: Anatomy







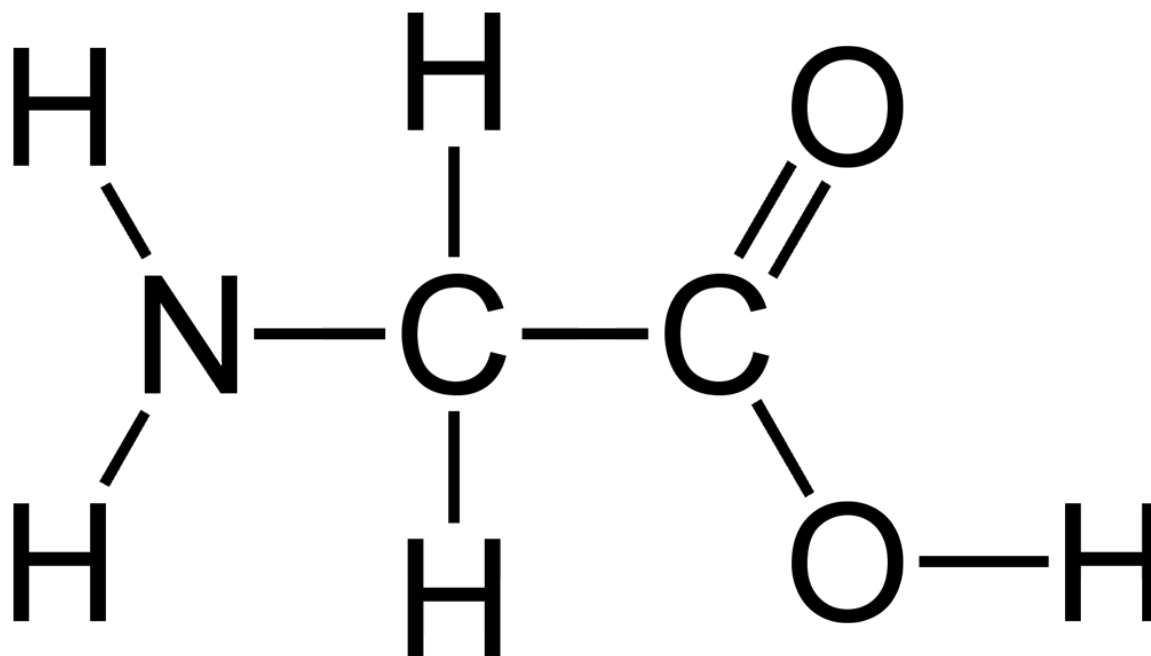
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Component	Whole Egg	White	Yolk
Weight	55g	38 g	17 g
Protein	6.6 g	3.9 g	2.7 g
Lipids	6 g	0 g	6 g
Calories	84 kcal	20 g	64 g

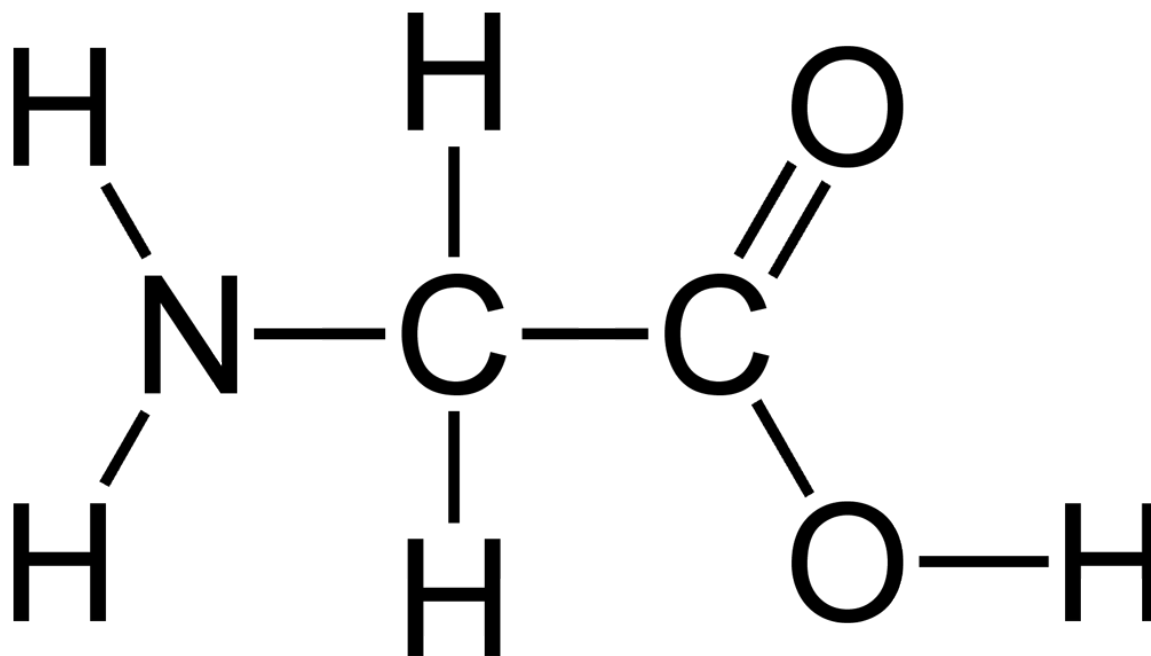
What are proteins and lipids?

What are **proteins** and lipids?

What is this?

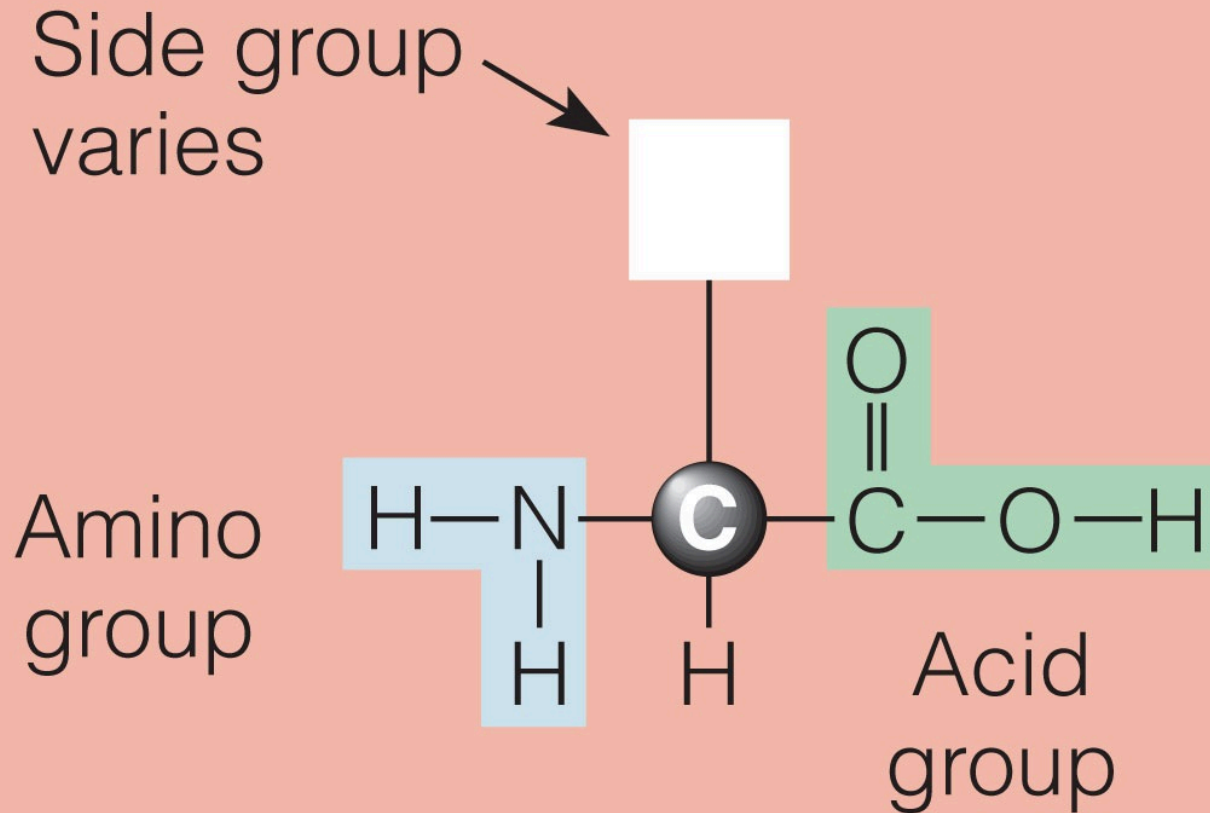


What is this?

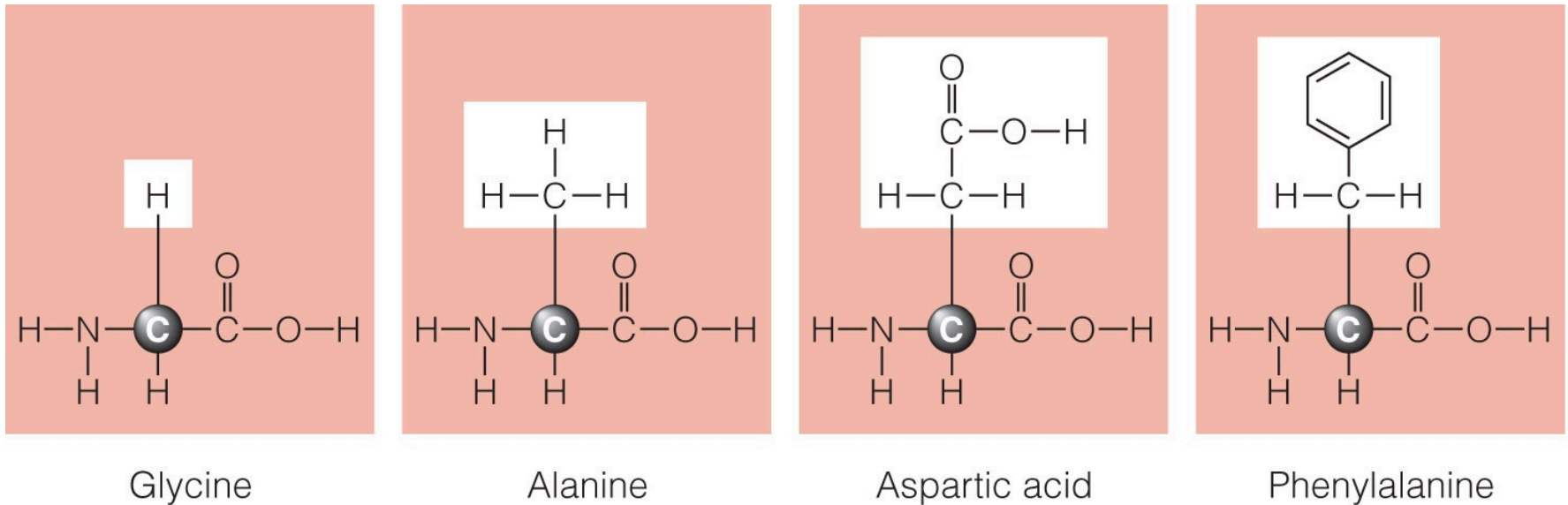


Glycine

Generic Amino Acid

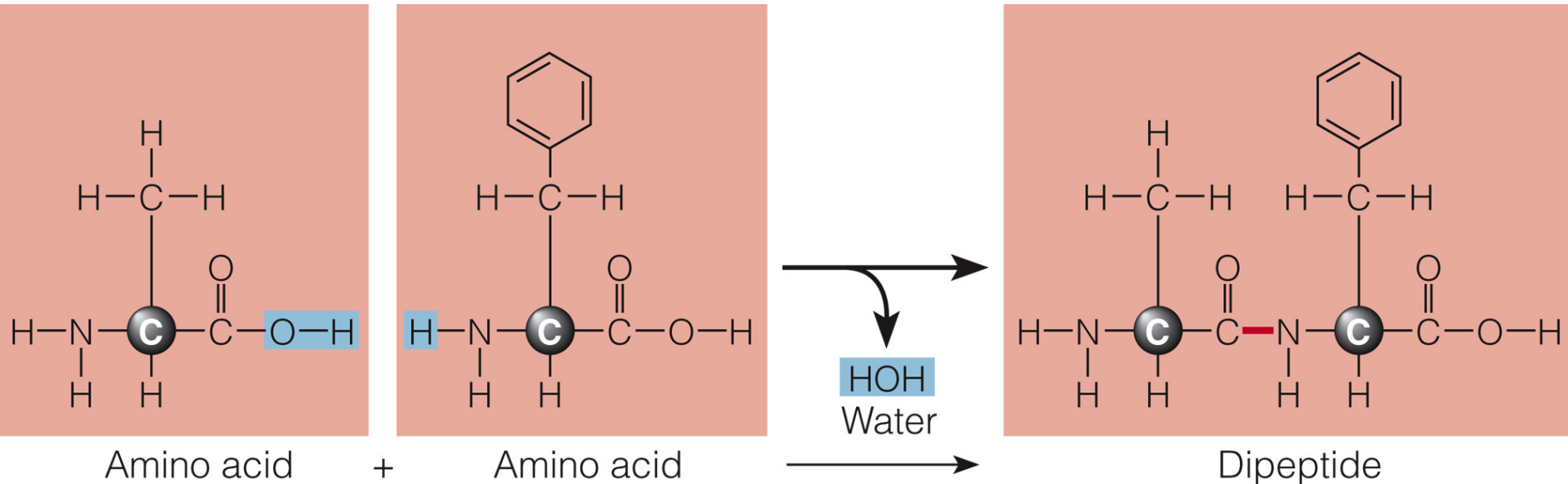


The side chains determine the properties of the amino acids



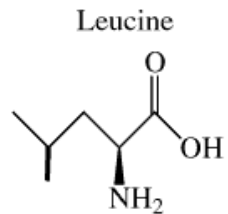
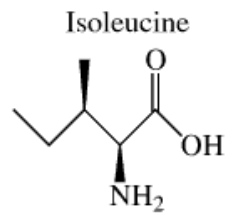
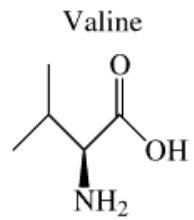
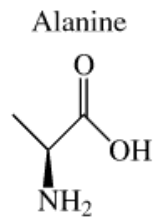
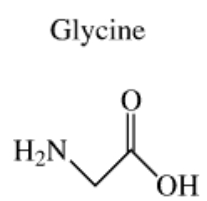
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Amino acid chains form peptides

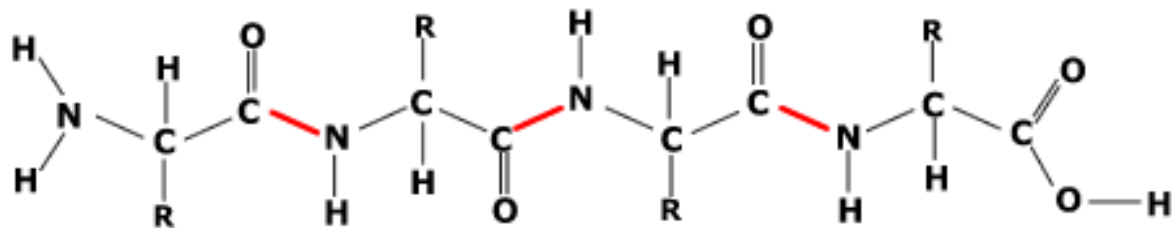


An OH group from the acid end of one amino acid and an H atom from the amino group of another join to form a molecule of water.

A peptide bond (highlighted in red) forms between the two amino acids, creating a dipeptide.

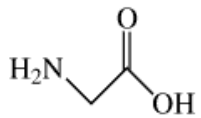


Amino acids

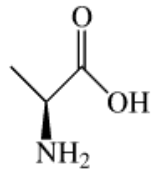


Polypeptide

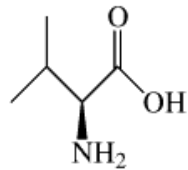
Glycine



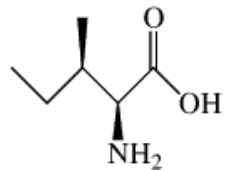
Alanine



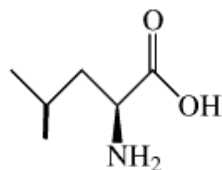
Valine



Isoleucine



Leucine



Amino acids

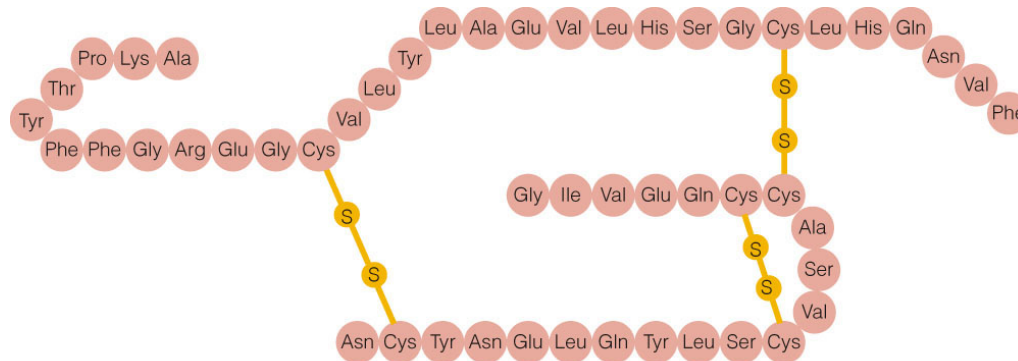
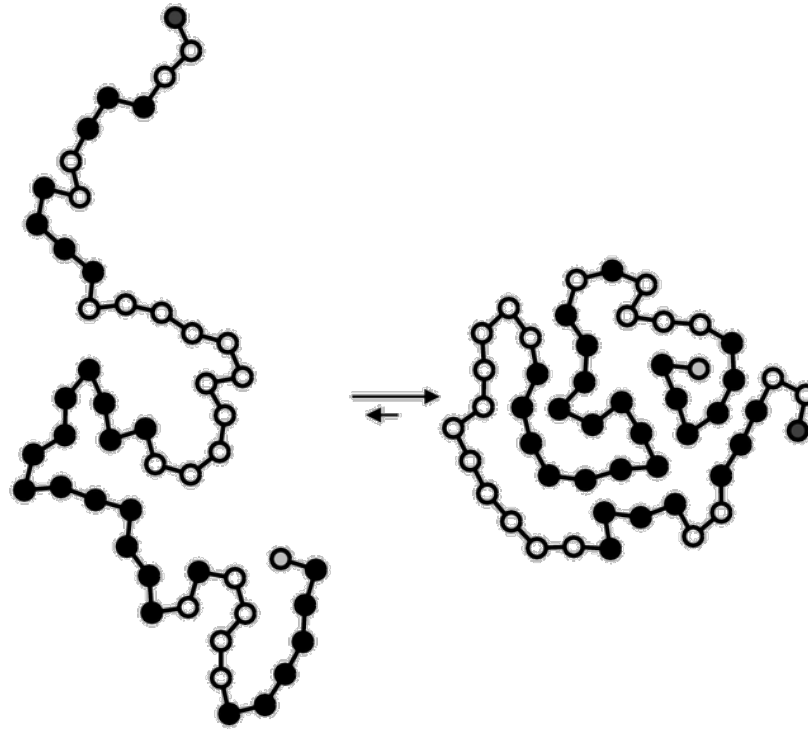
Amino Acids and Protein Folding

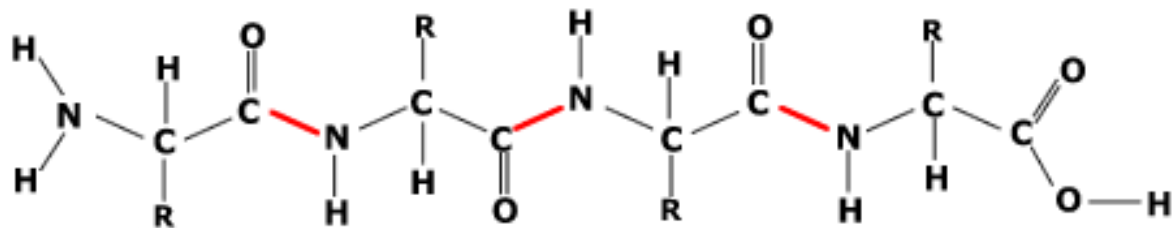
polar
(hydrophilic)
LOVES WATER

arginine
asparagine
aspartic acid
glutamic acid
glutamine
histidine
lysine
serine
threonine
tyrosine

nonpolar
(hydrophobic)
HATES WATER

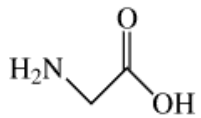
alanine
cysteine
glycine
isoleucine
leucine
methionine
phenylalanine
proline
tryptophan
valine



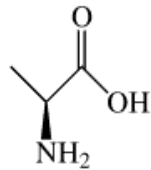


Polypeptide

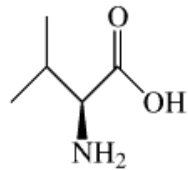
Glycine



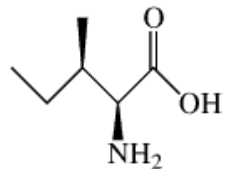
Alanine



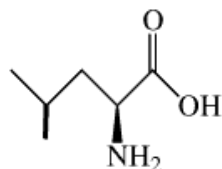
Valine



Isoleucine

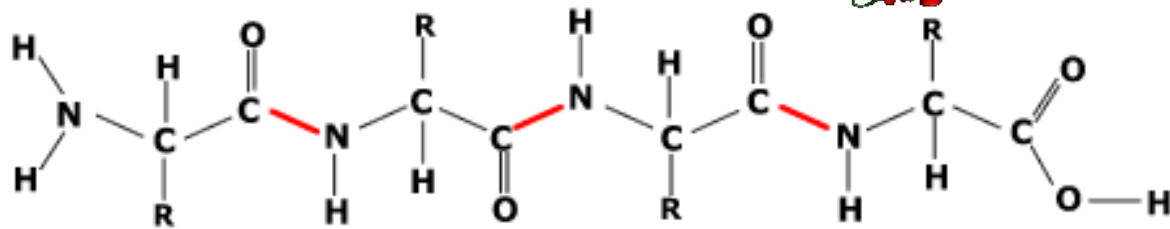
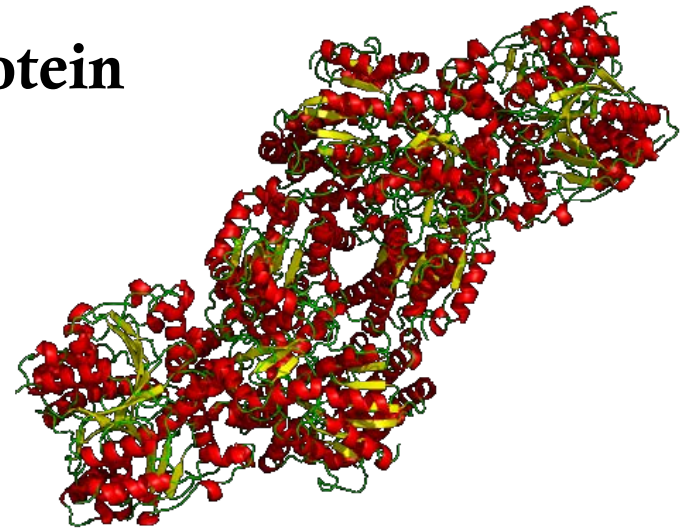


Leucine



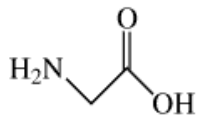
Amino acids

Protein

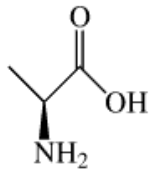


Polypeptide

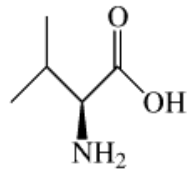
Glycine



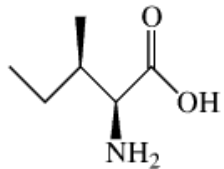
Alanine



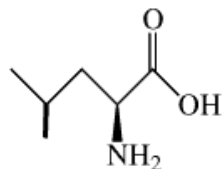
Valine



Isoleucine



Leucine

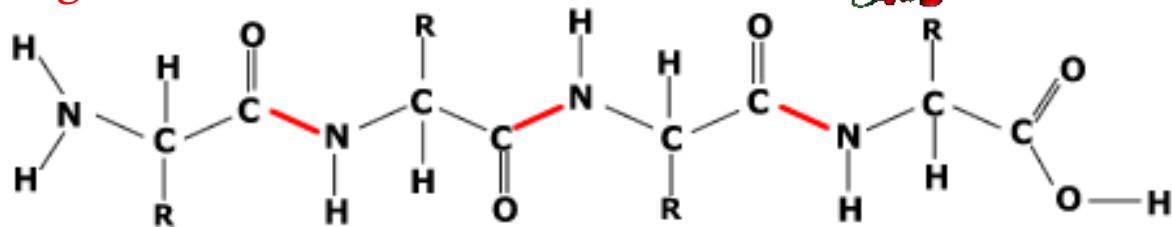
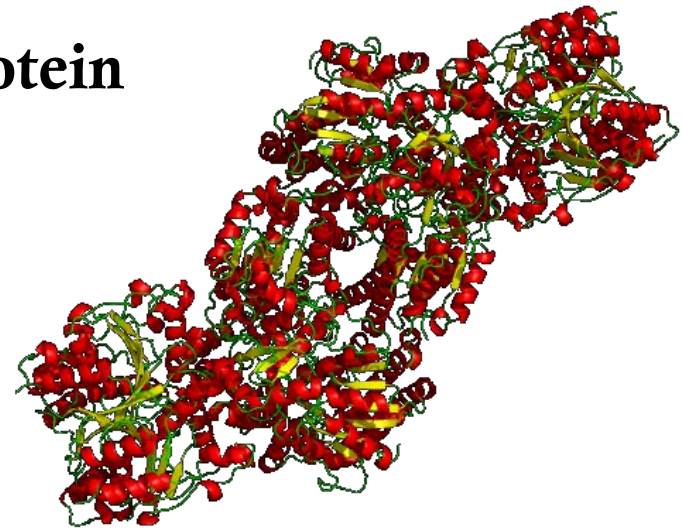


Amino acids



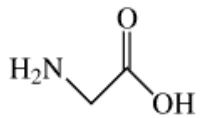
6 google image for “weird human”

Protein

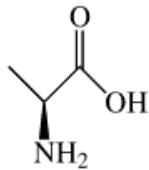


Polypeptide

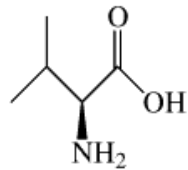
Glycine



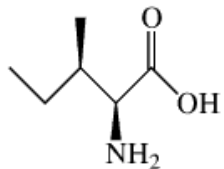
Alanine



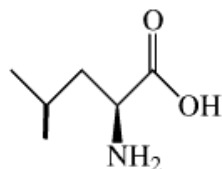
Valine



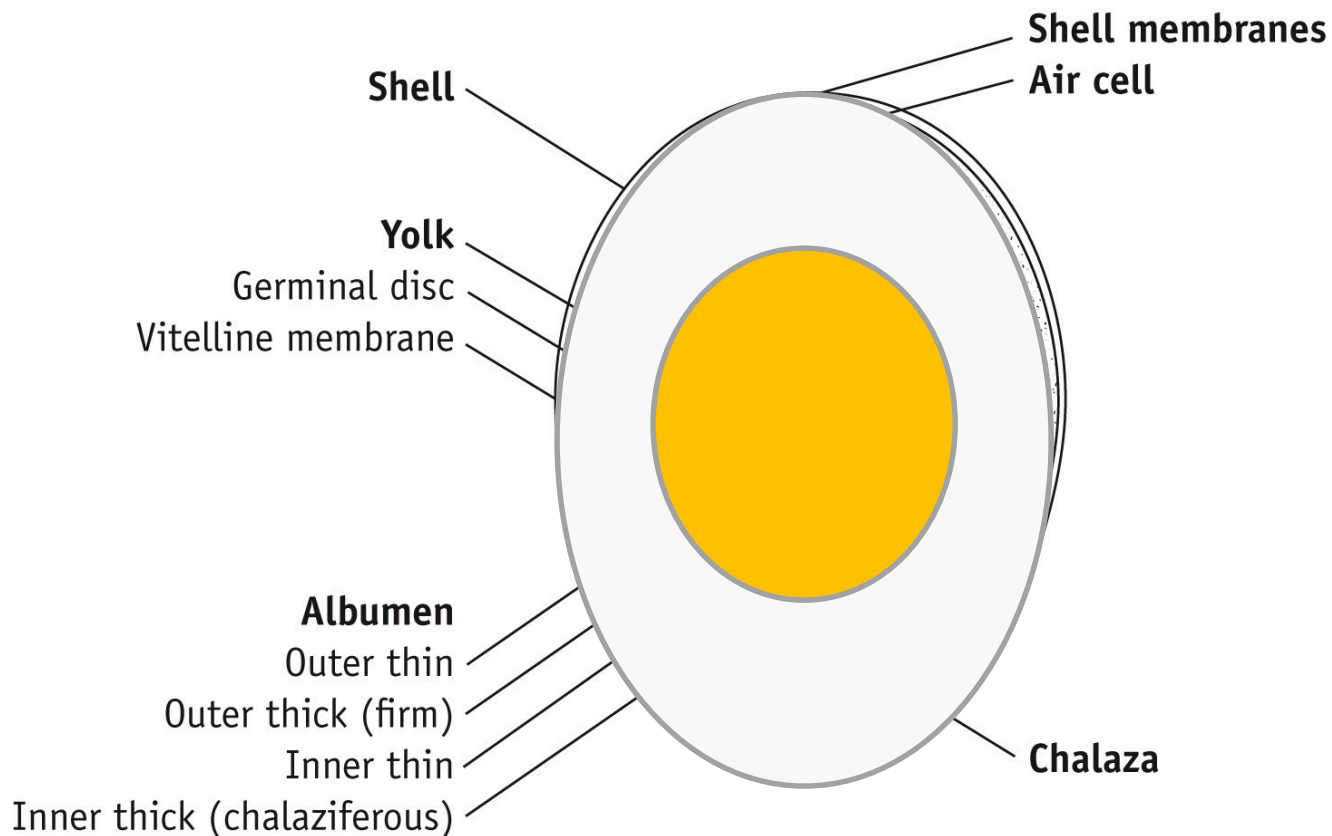
Isoleucine



Leucine



Amino acids



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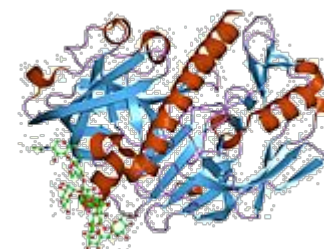
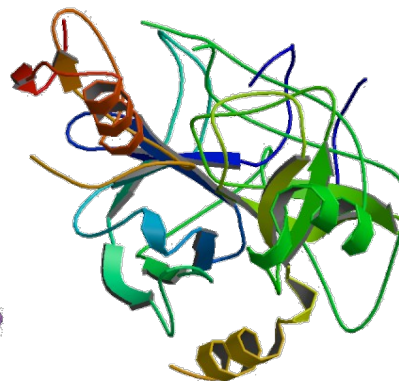
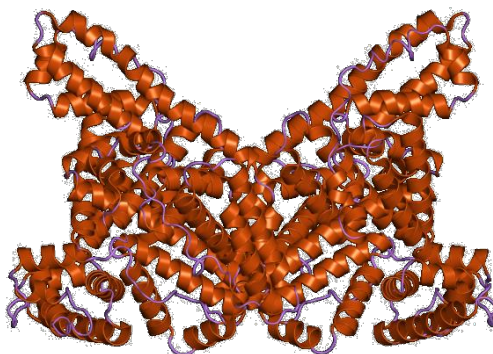
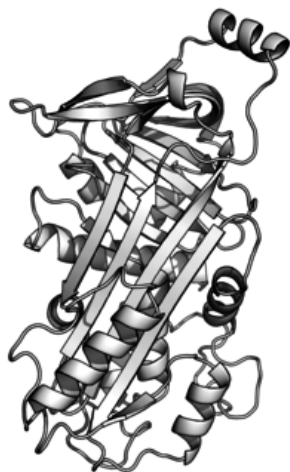
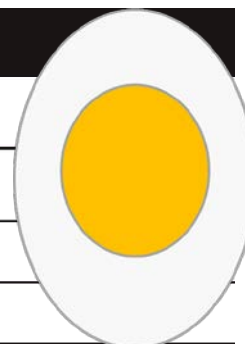


TABLE 12-1 Protein Content of Egg White

Protein	Amount (%)*	Properties
Ovalbumin	54	Denatures easily
Conalbumin	13	Antimicrobial, complexes iron
Ovomucoid	11	Inhibits enzyme (trypsin)
Unidentified	8	Mainly globulins
Lysozyme	3.5	Antimicrobial
Ovomucin	1.5	Viscous, reacts with viruses
Flavoprotein	0.8	Binds riboflavin
Proteinase inhibitor	0.1	Inhibits enzyme (bacterial proteinase)
Avidin	0.05	Binds the B vitamin biotin (raw egg white only)



*Expressed as a percentage of the total protein content of the egg white.

What are proteins and **lipids**?

The Family of Lipids

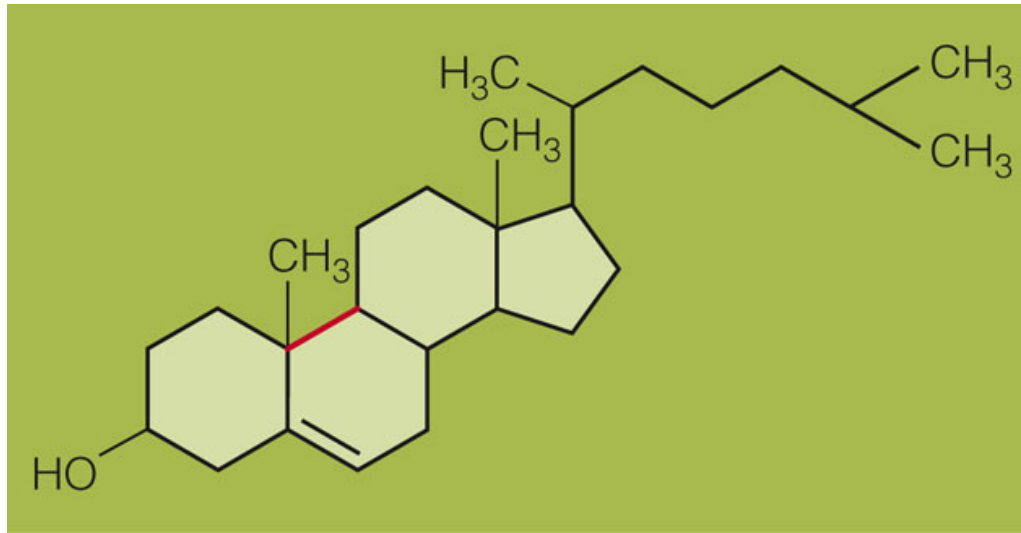
HYDROPHOBIC (HATES WATER):

- Sterols
- Triglycerides

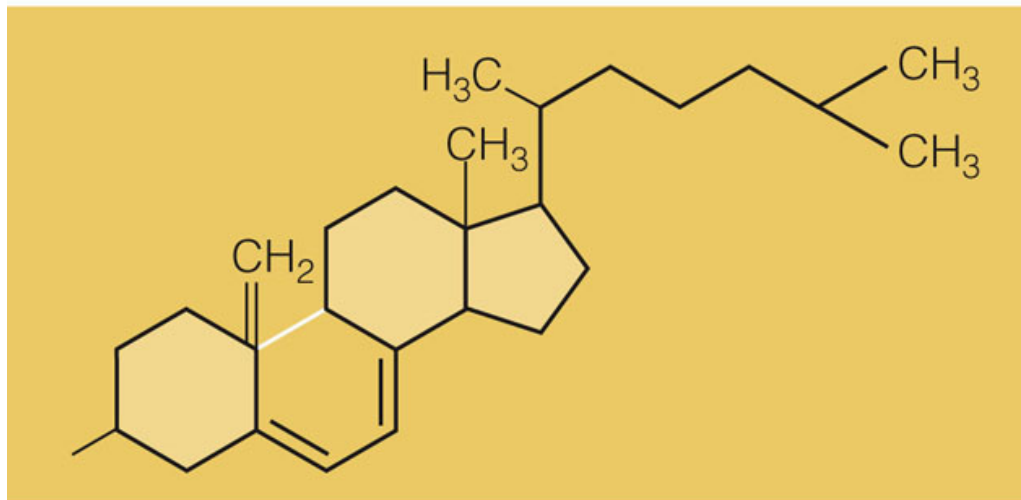
Part HYDROHOBIC, Part HYDROPHILIC (LOVE HATE WITH WATER):

- Phospholipids

Sterols HYDROPHOBIC (HATES WATER):

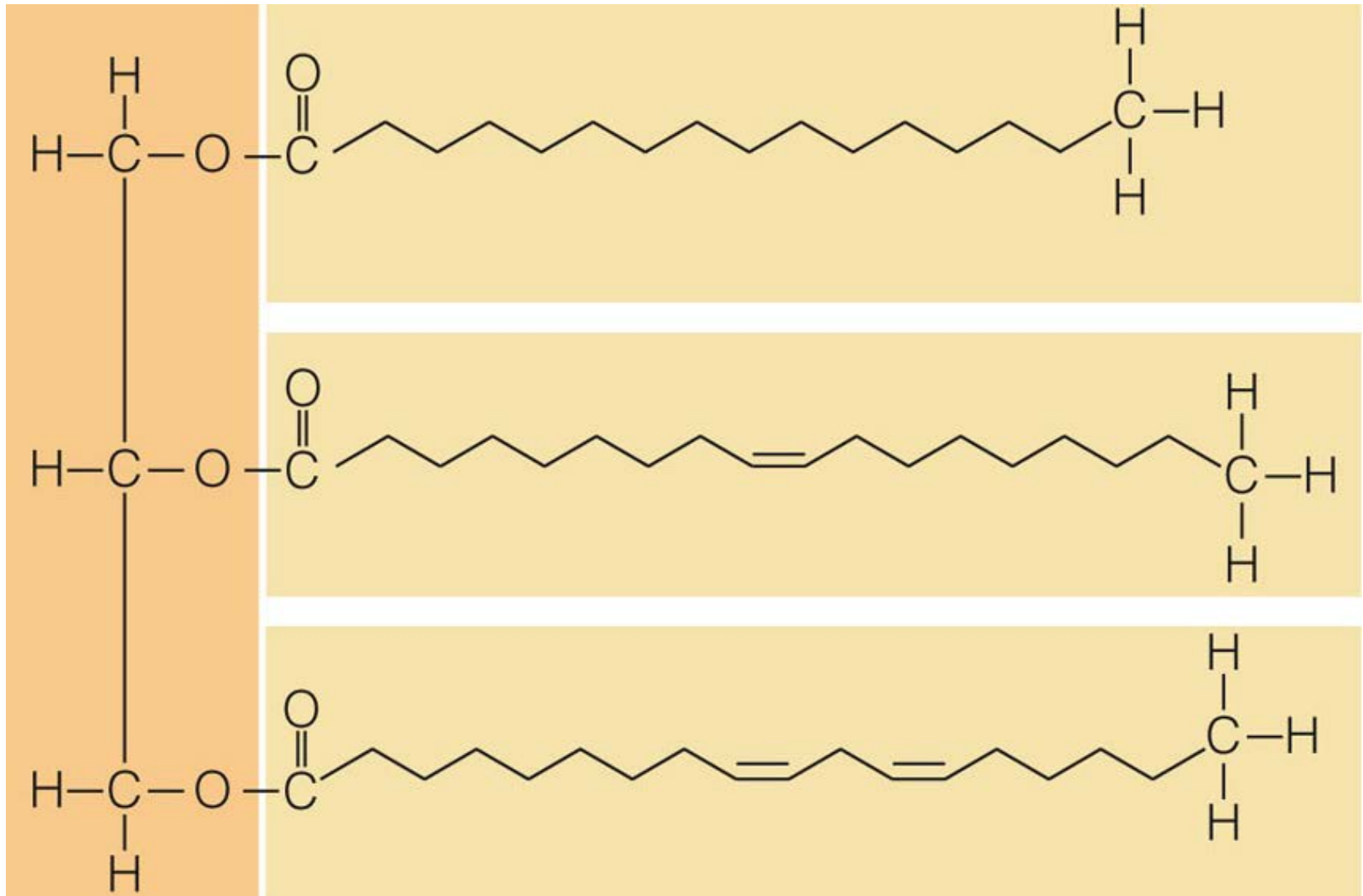


Cholesterol

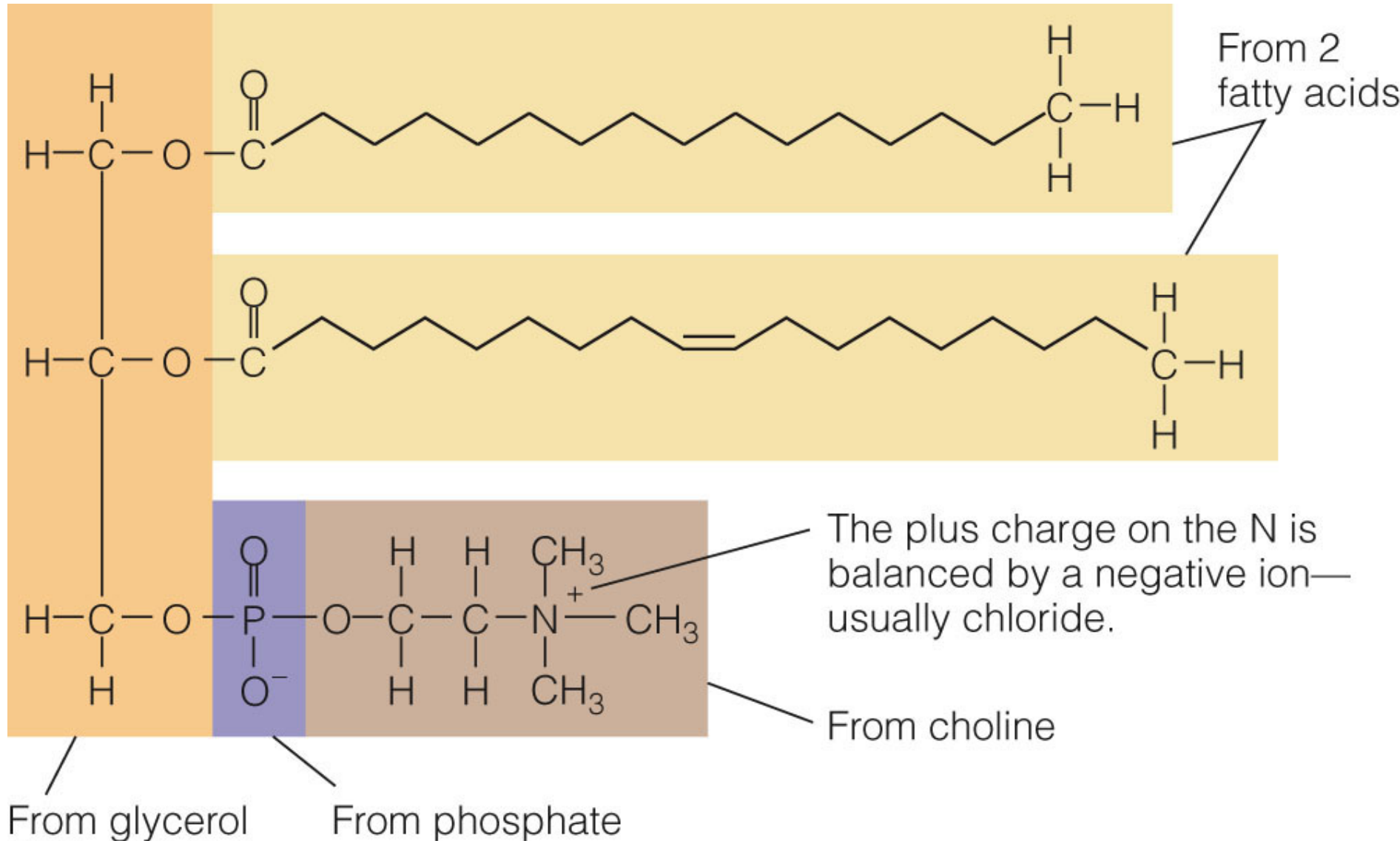


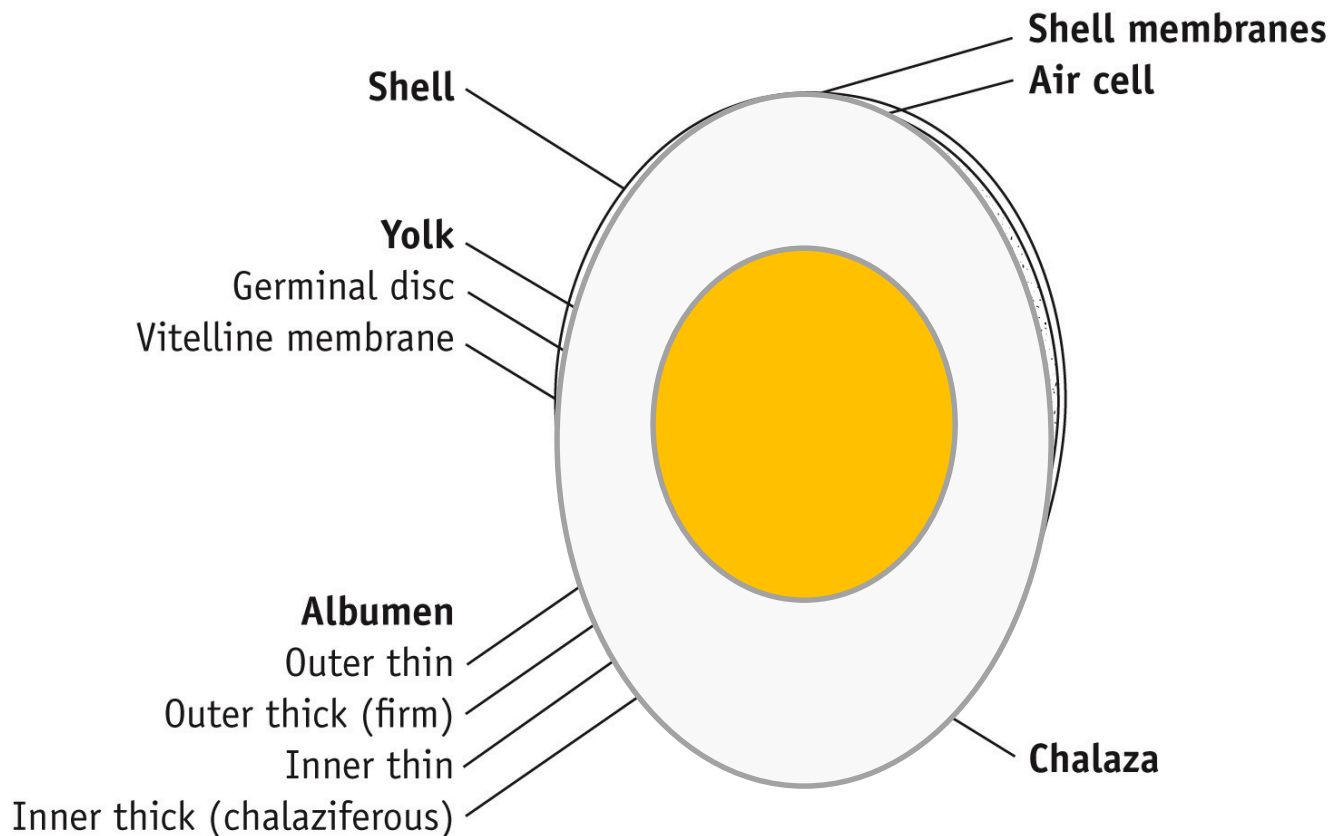
Vitamin D₃

A generic triglyceride HYDROPHOBIC (HATES WATER):



Phospholipids: (LOVE HATE WITH WATER):

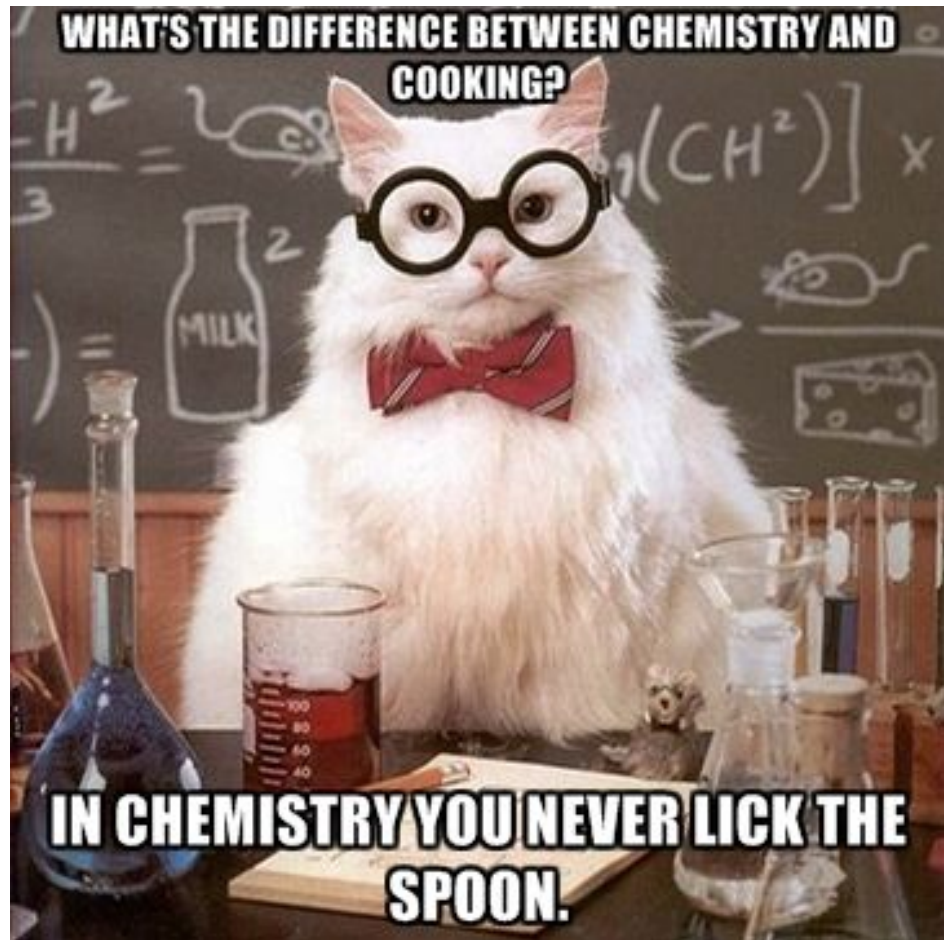




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What's Cooking?



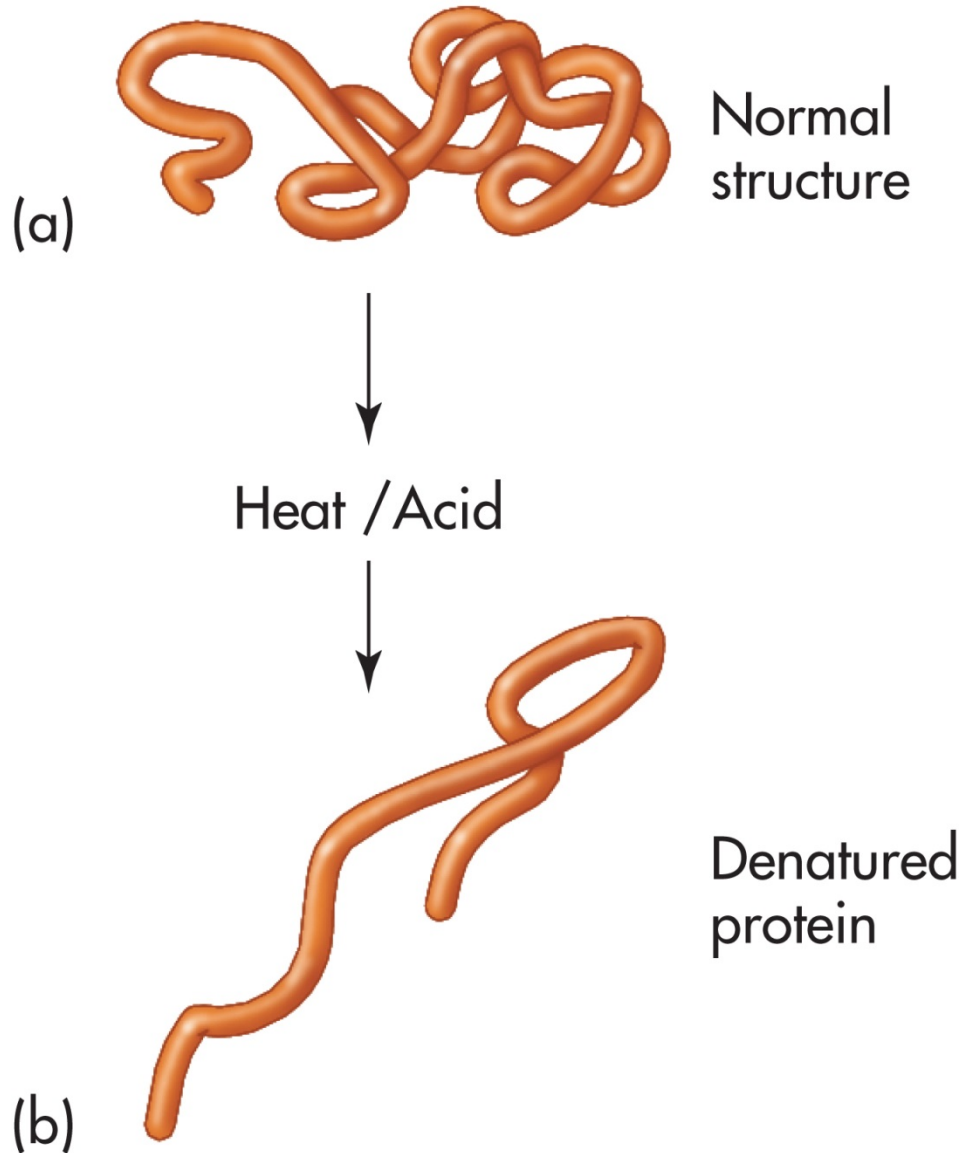
Denaturing proteins: Just apply heat



© Matthew Farruggio

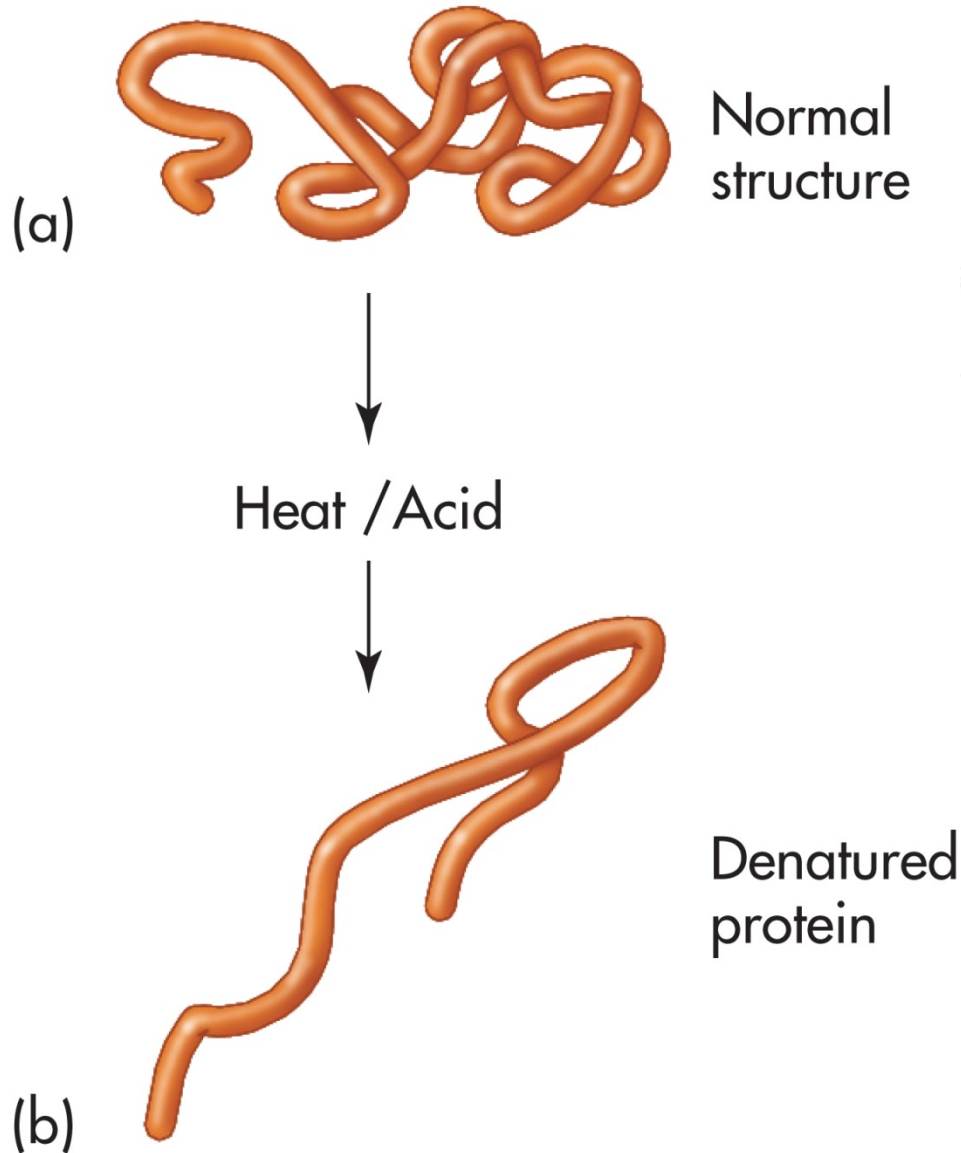
Denaturation of Proteins

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Denaturation of Proteins

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“Renaturation” of Proteins

Shear-Stress-Mediated Refolding of Proteins from Aggregates and Inclusion Bodies

CHEMBIOCHEM
Communications

Tom Z. Yuan,^[a] Callum F. G. Ormonde,^[b] Stephan T. Kudlacek,^[c] Sameeran Kunche,^[d]
Joshua N. Smith,^[a] William A. Brown,^[c] Kaitlin M. Pugliese,^[c] Tivoli J. Olsen,^[c] Mariam Iftikhar,^[a]
Colin L. Raston,^{*[e]} and Gregory A. Weiss^{*[a, c]}



“Renaturation” of Proteins

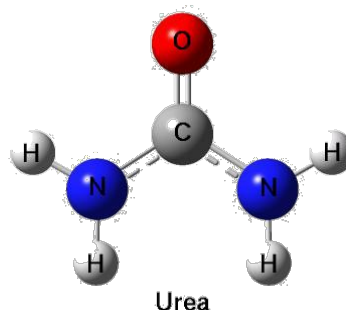
Shear-Stress-Mediated Refolding of Proteins from Aggregates and Inclusion Bodies

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Colin L. Raston,^{*,[e]} and Gregory A. Weiss^{*,[a, c]}



+



“Renaturation” of Proteins

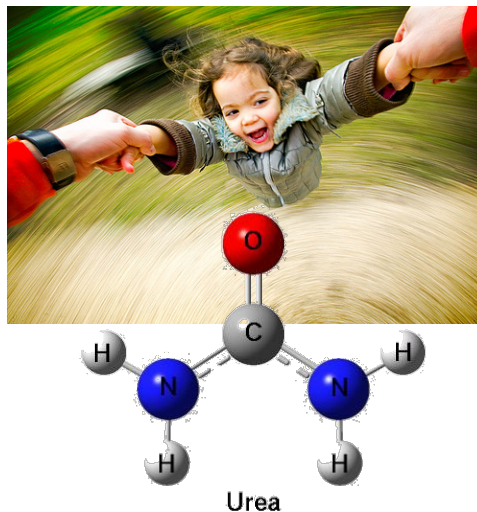
Shear-Stress-Mediated Refolding of Proteins from Aggregates and Inclusion Bodies

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Colin L. Raston,^{*,[e]} and Gregory A. Weiss^{*,[a, c]}



+

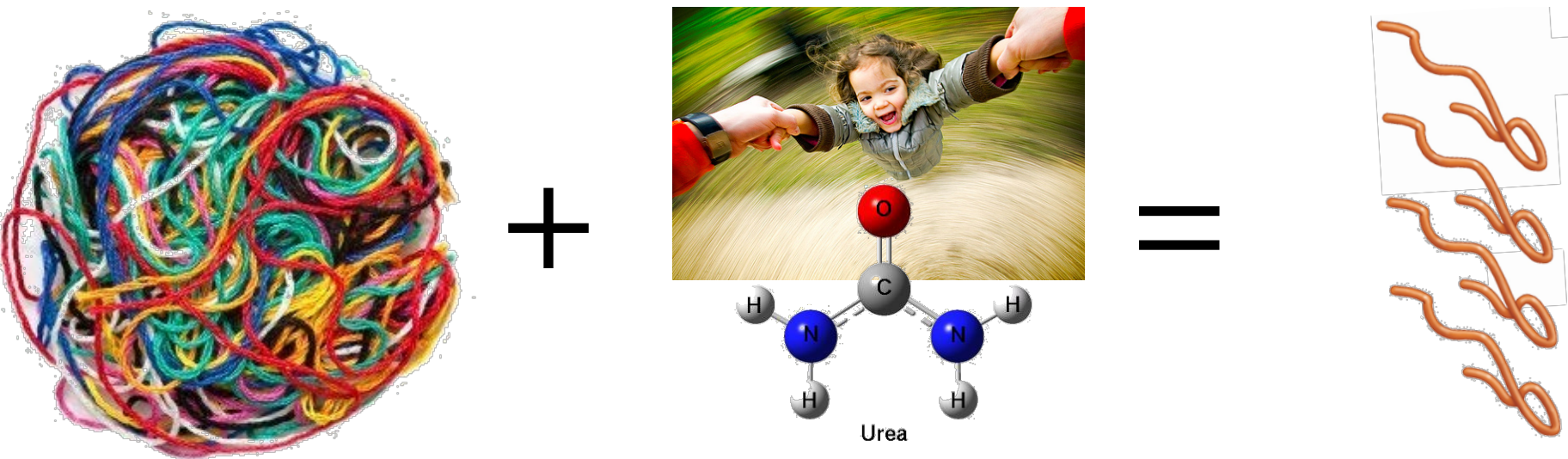


“Renaturation” of Proteins

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Colin L. Raston,^{*,[e]} and Gregory A. Weiss^{*,[a, c]}

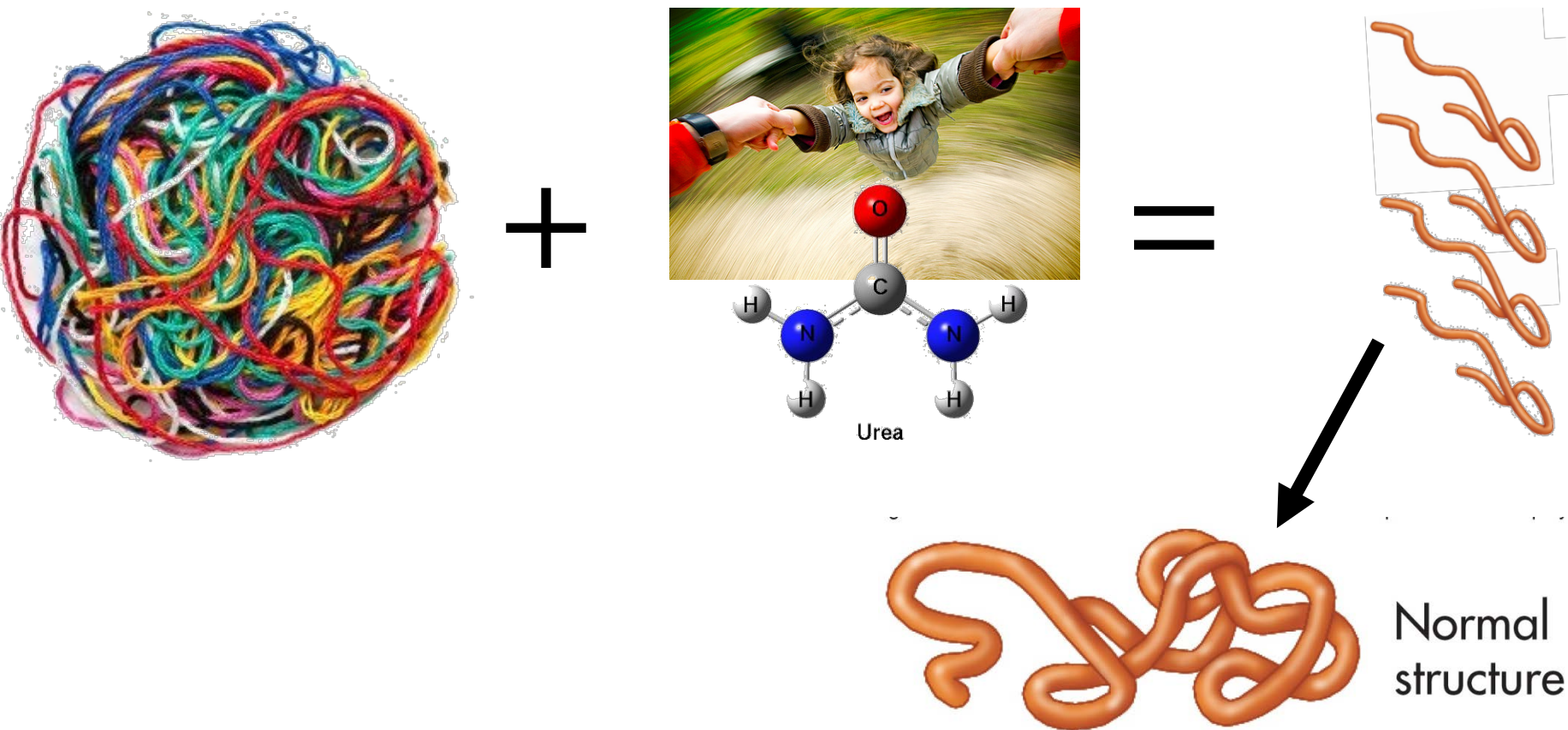


“Renaturation” of Proteins

Shear-Stress-Mediated Refolding of Proteins from Aggregates and Inclusion Bodies

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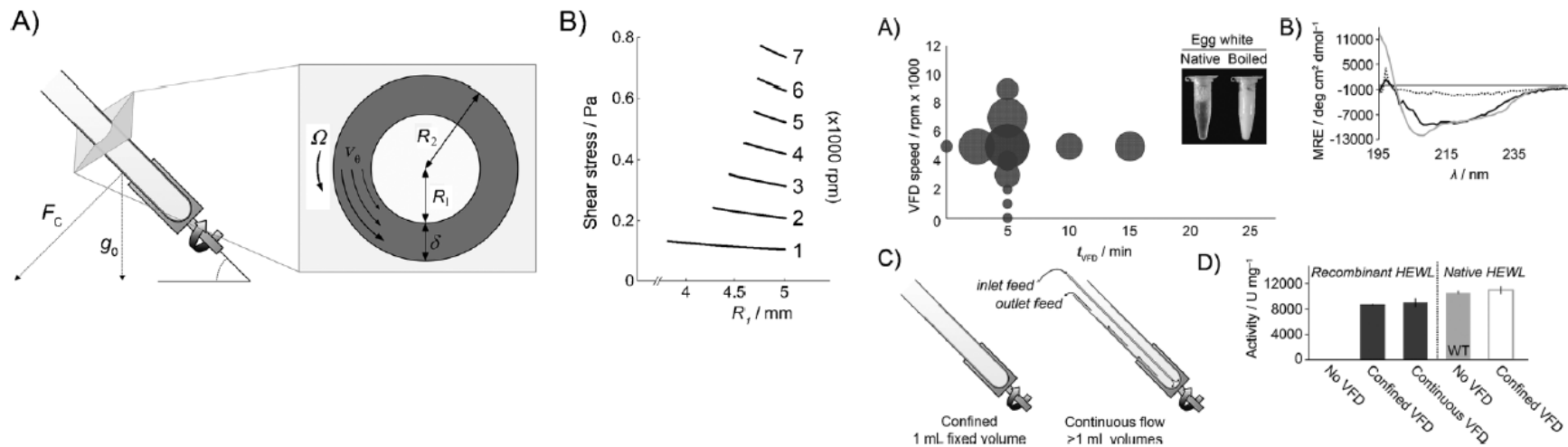


“Renaturation” of Proteins

Shear-Stress-Mediated Refolding of Proteins from Aggregates and Inclusion Bodies

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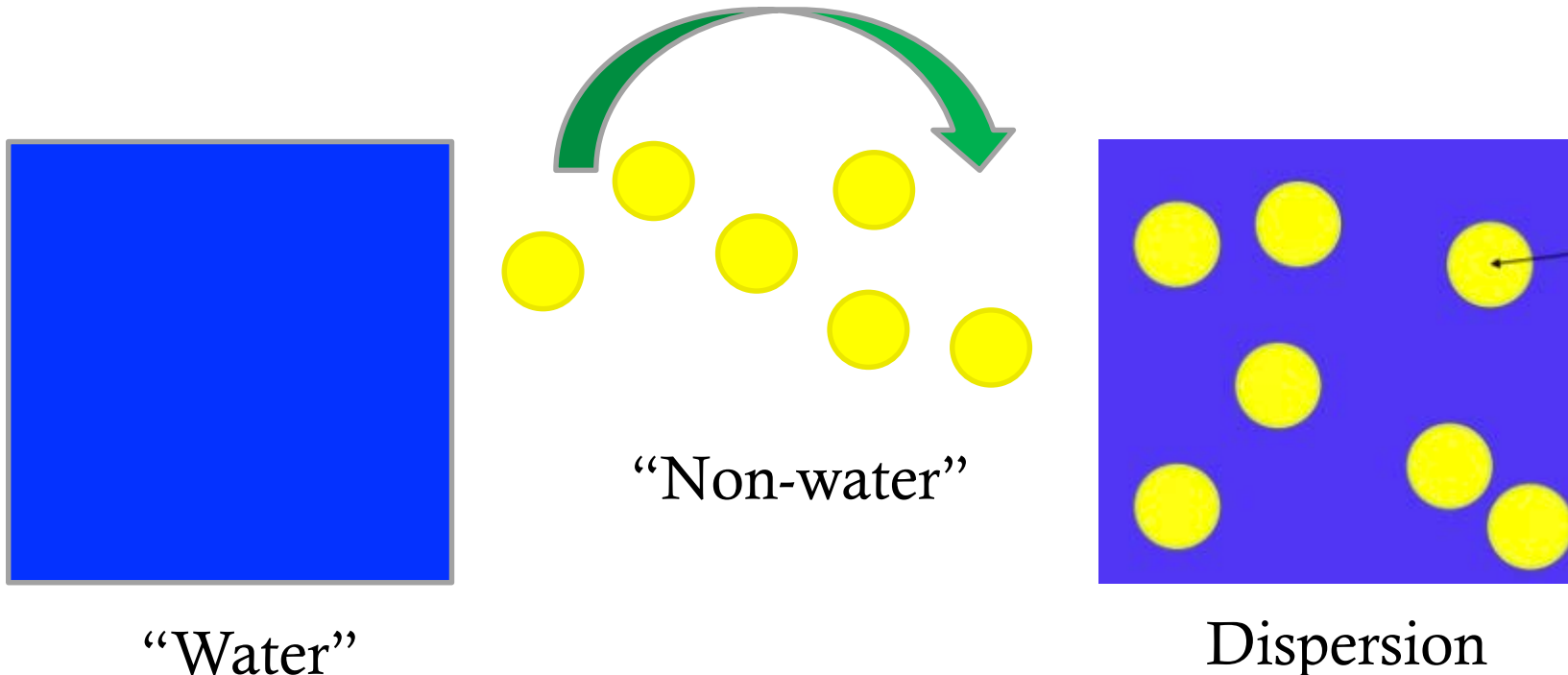
Tom Z. Yuan,^[a] Callum F. G. Ormonde,^[b] Stephan T. Kudlacek,^[c] Sameeran Kunche,^[d]
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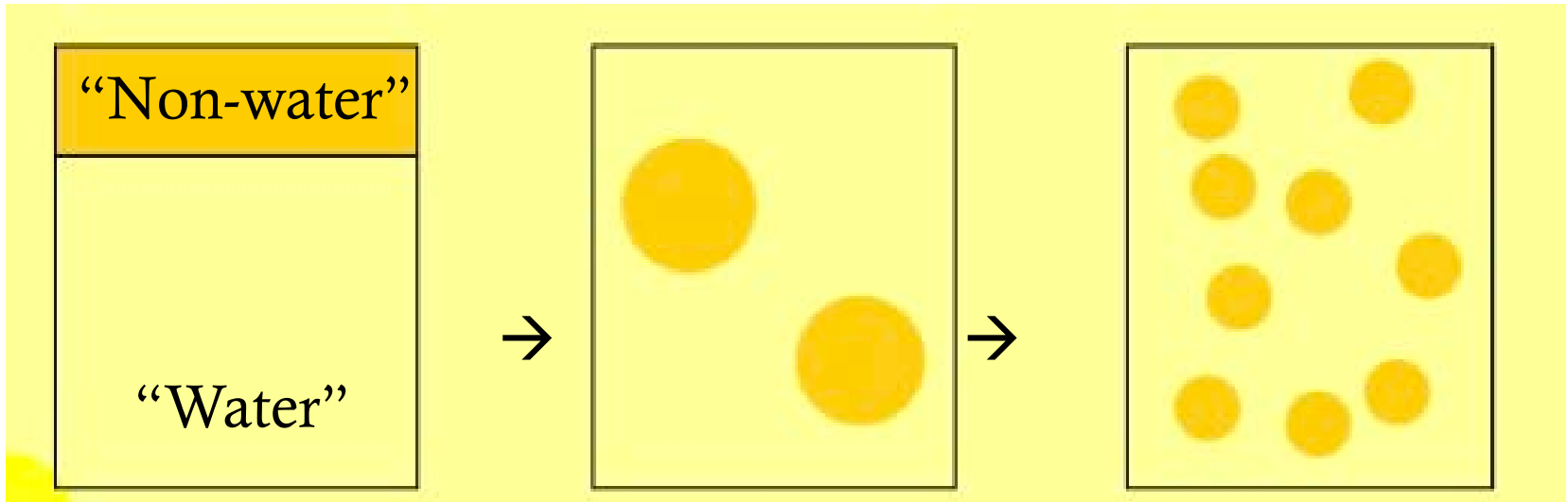
Thickening liquids:
It's all mixed up

How do you make “water” less “watery”?

Cram it full of “non-water”!



The more you mix it, the less “watery” it gets



Two things happening here:

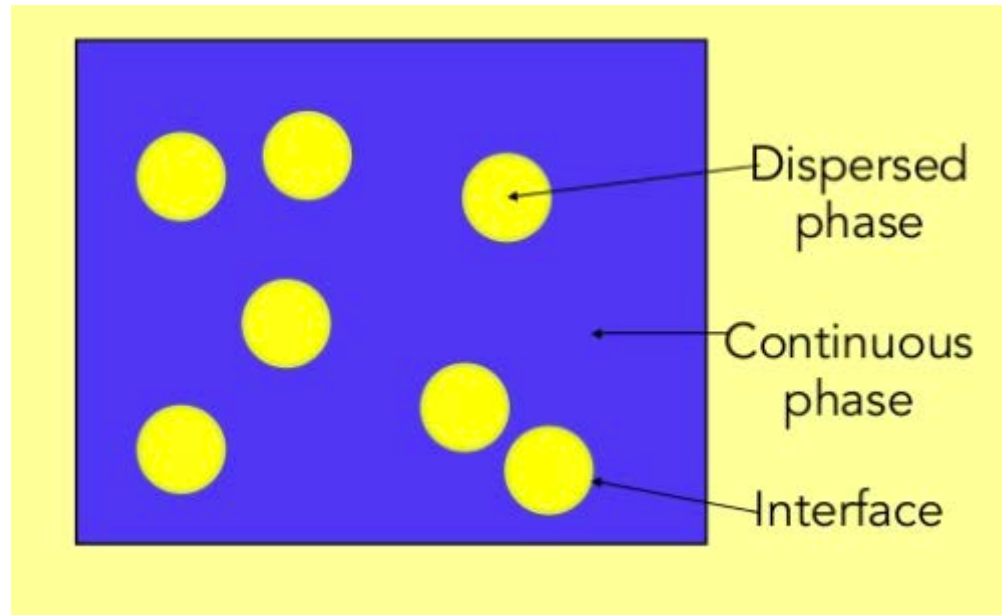
Droplet size of “non-water” decreases &

Surface area contact between “water” & “non-water” increases

Types of “non-water”:

- Air bubbles
- Oil droplets
- Plant or animal particles

The technical terms



Types of Dispersions

		CONTINUOUS PHASE		
DISPERSED PHASE		SOLID	LIQUID	GAS
	SOLID	Some glasses	Sol	Smoke
	LIQUID	Gel	Emulsion	Aerosol
	GAS	Solid foam	Foam	

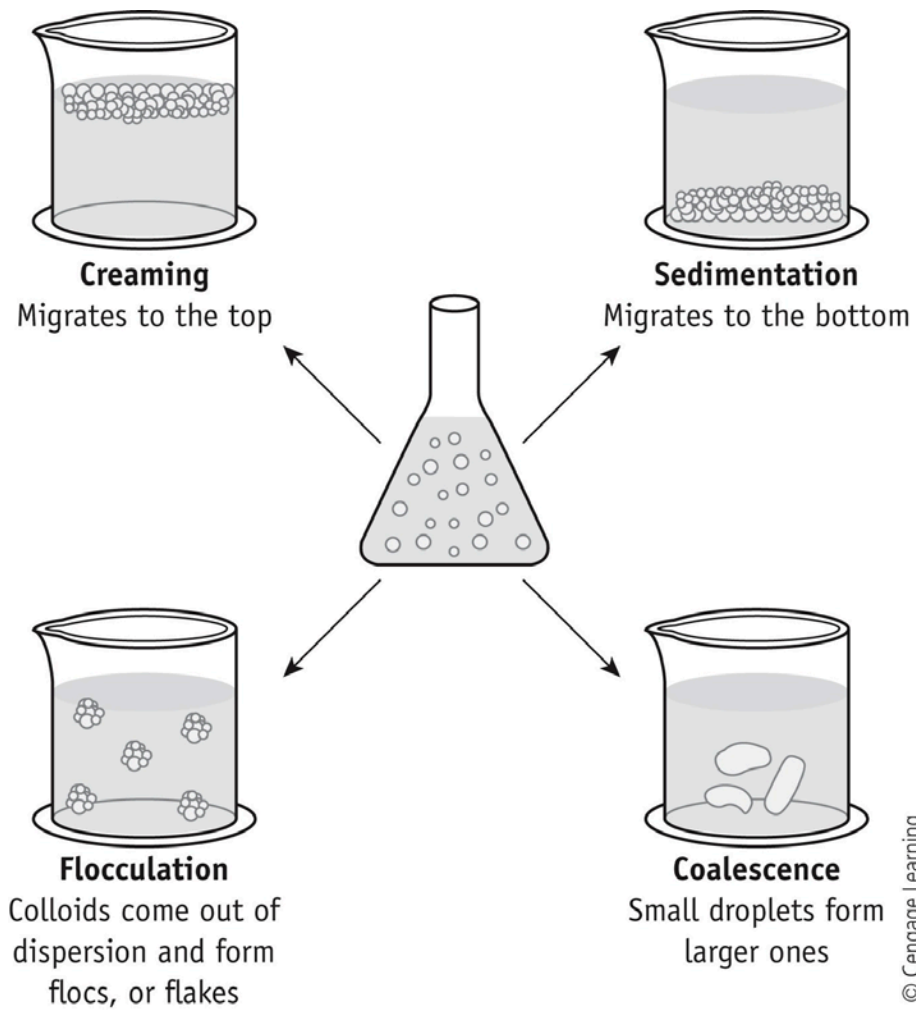
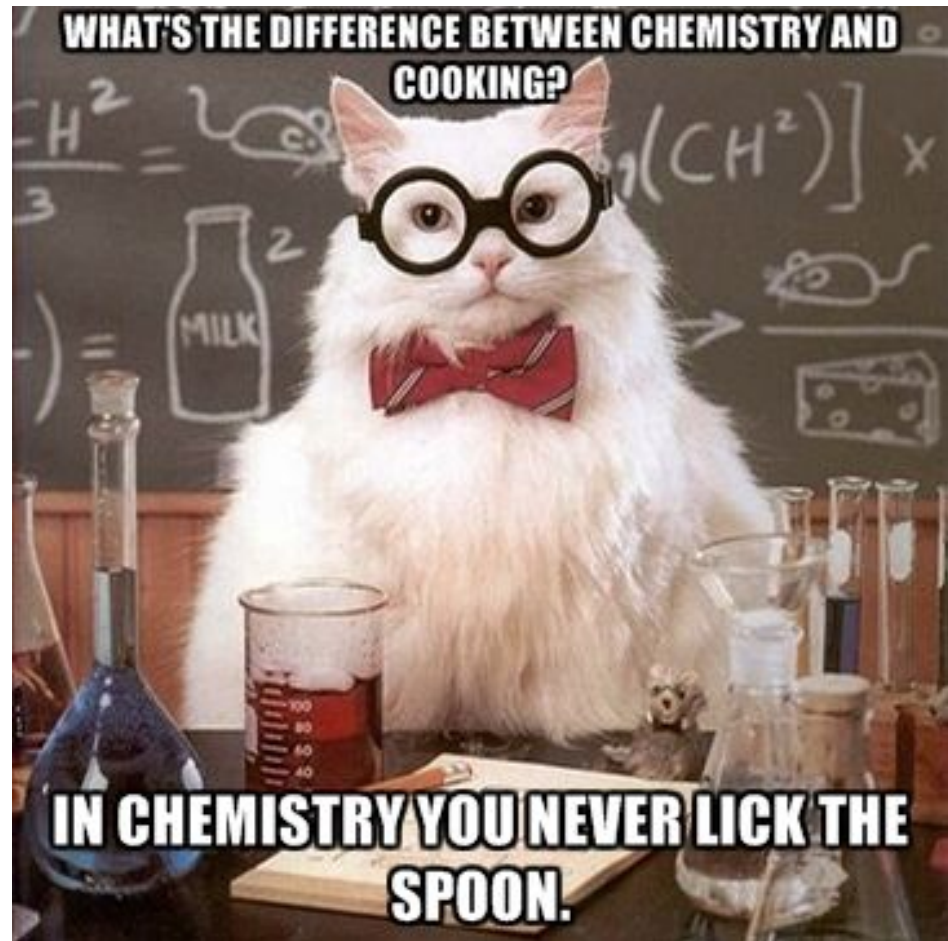


Figure 3-7 p38

What's Cooking?



Thickening by Adding Air:

FOAMS





(a) Foamy



(b) Soft peaks



(c) Stiff peaks

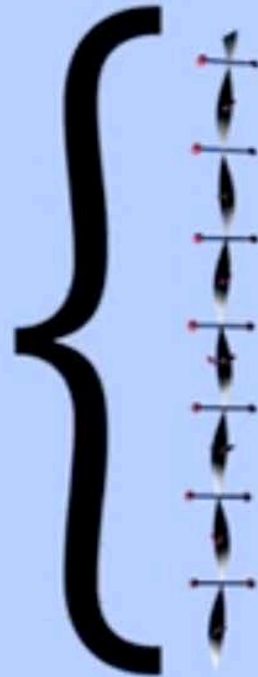


(d) Dry peaks

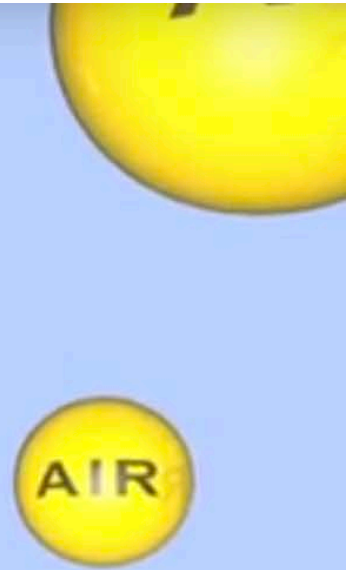
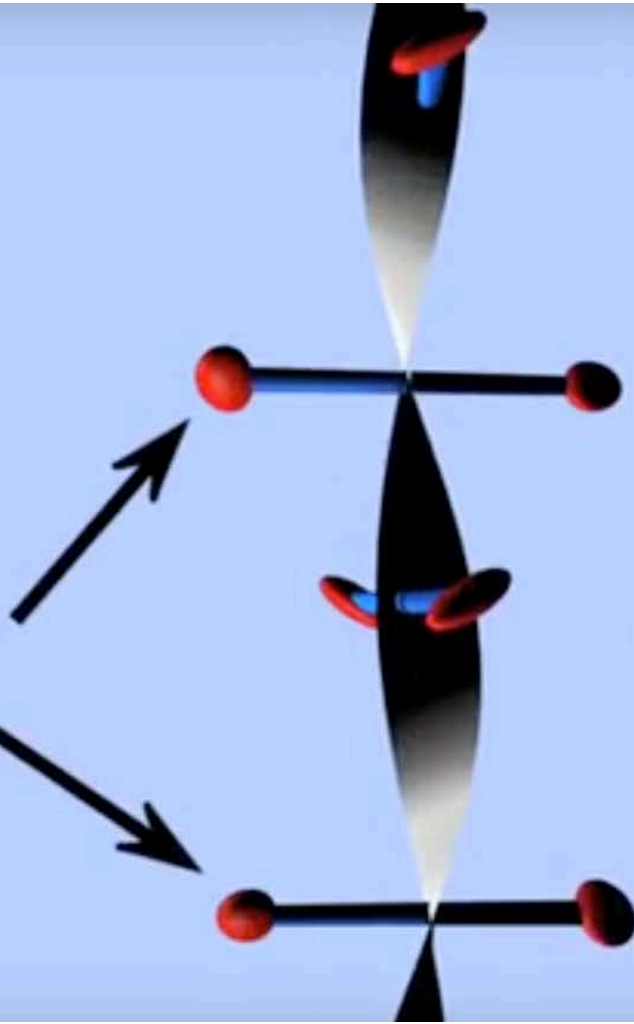
Digital Works

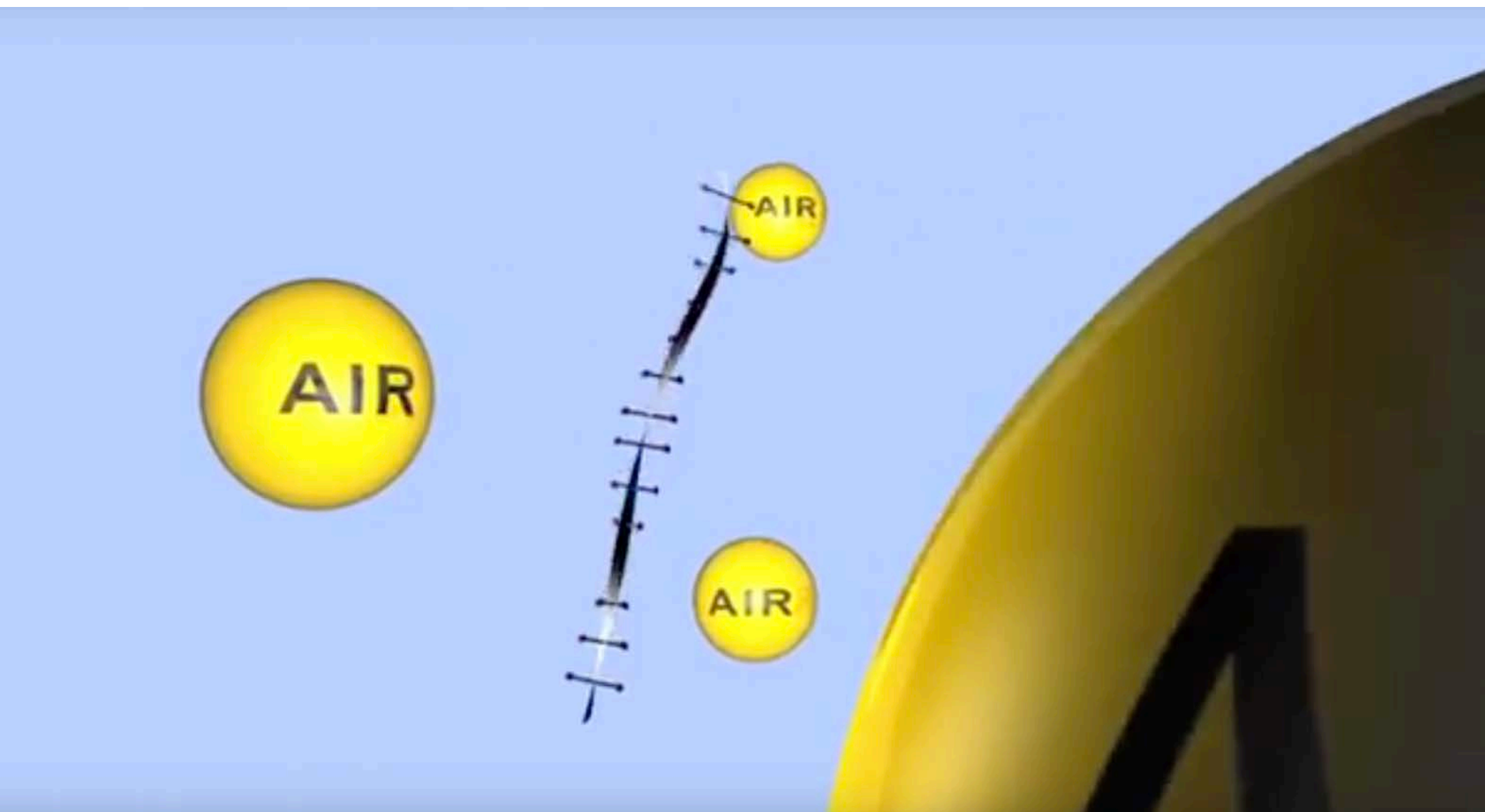
Figure 12-8 p258

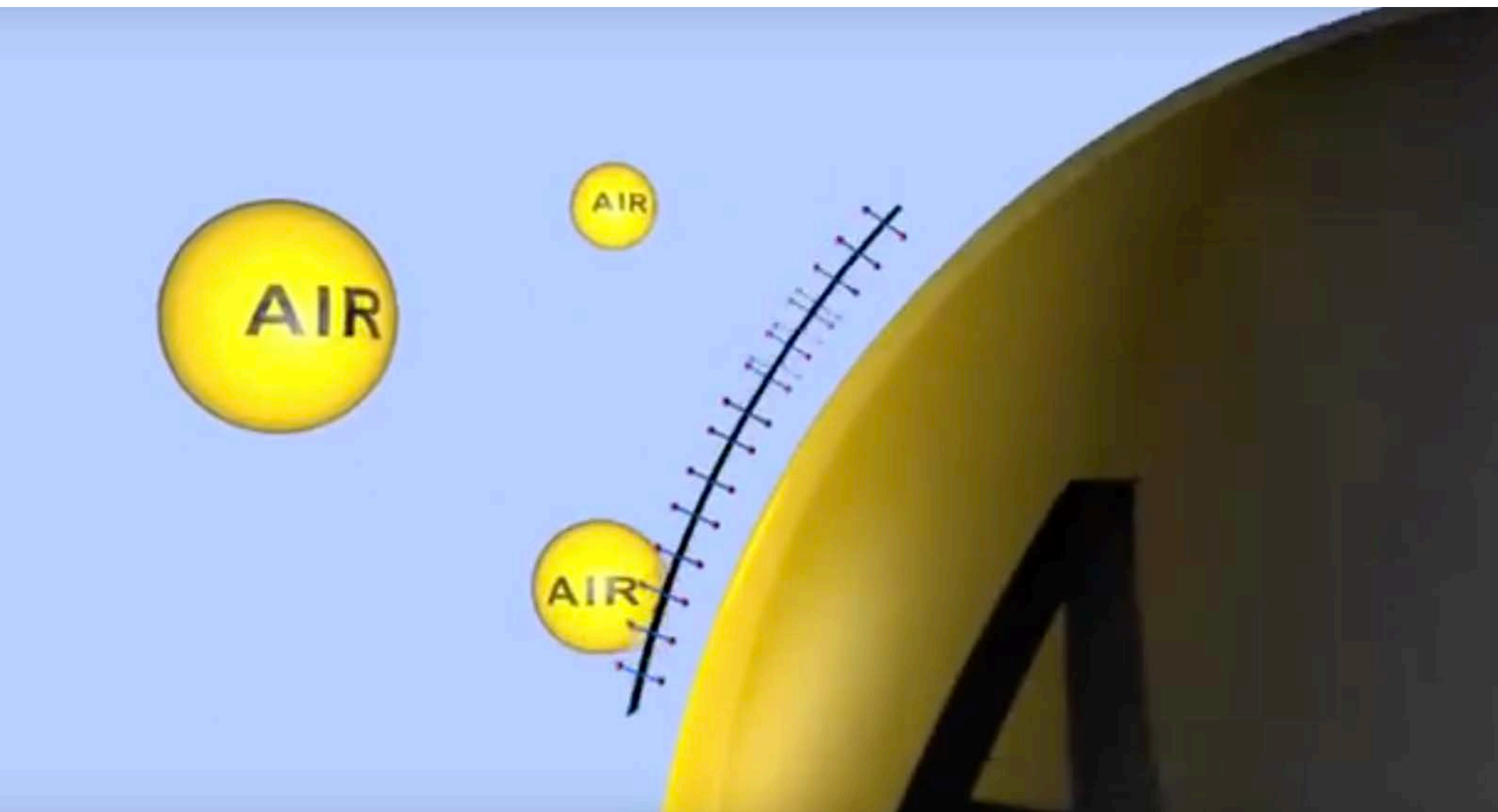
EGG PROTEIN



**AMINO
ACIDS**

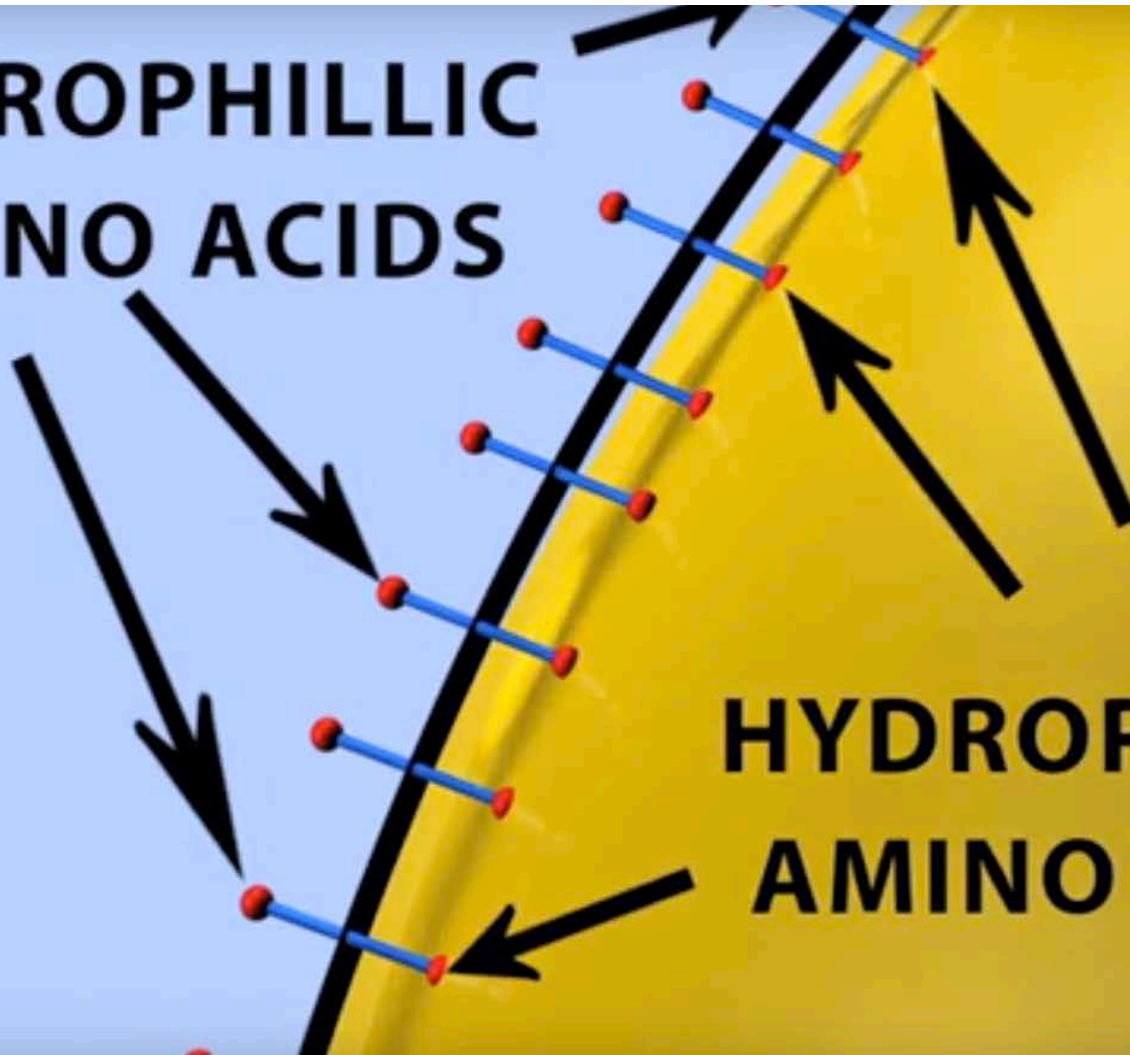




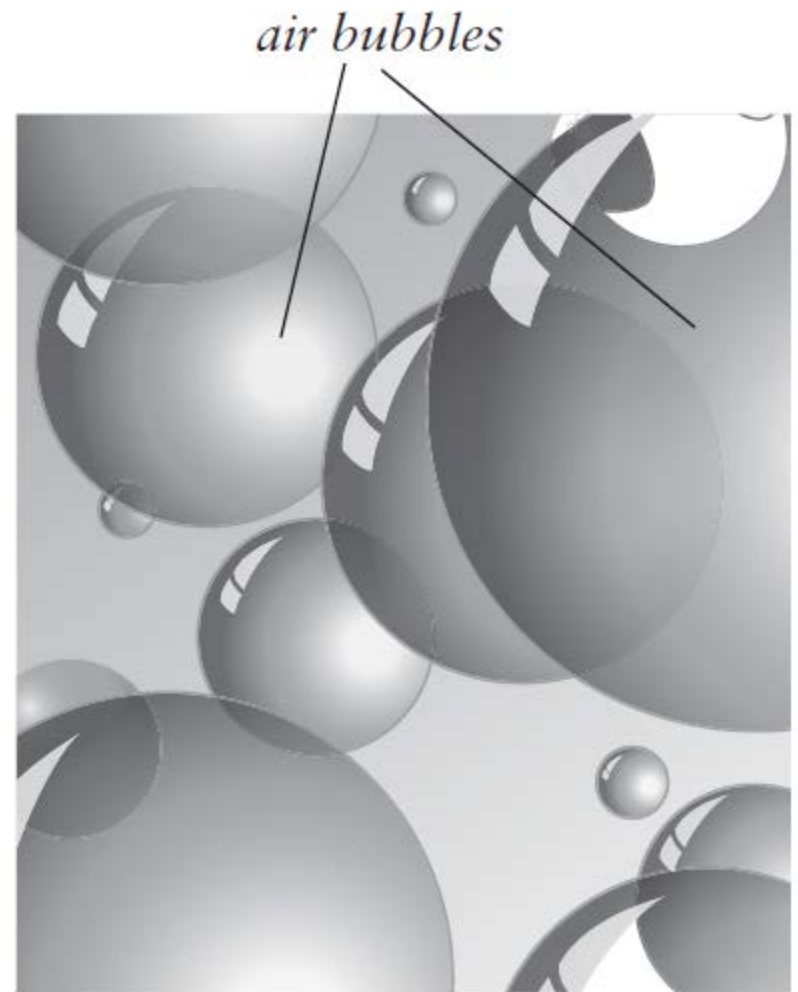


**HYDROPHILIC
AMINO ACIDS**

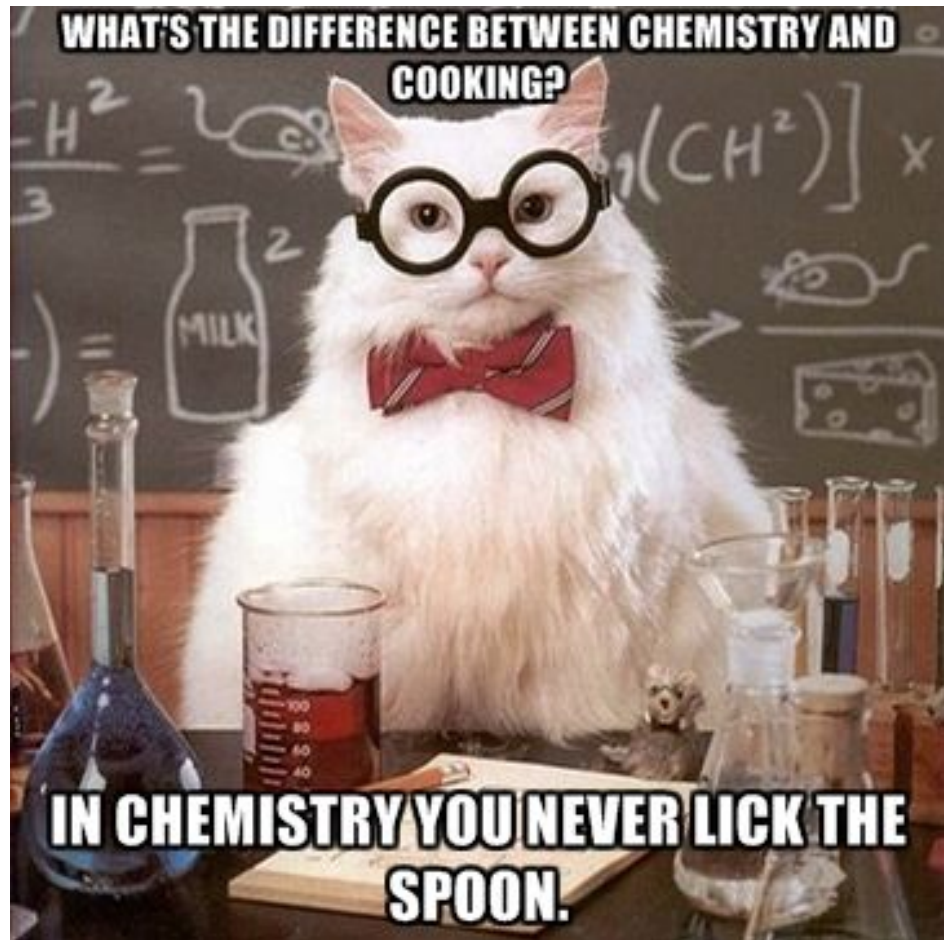
**HYDROPHOBIC
AMINO ACIDS**



Thickening with air:



What's Cooking?



Thickening by Combining Oil and Water: EMULSIONS



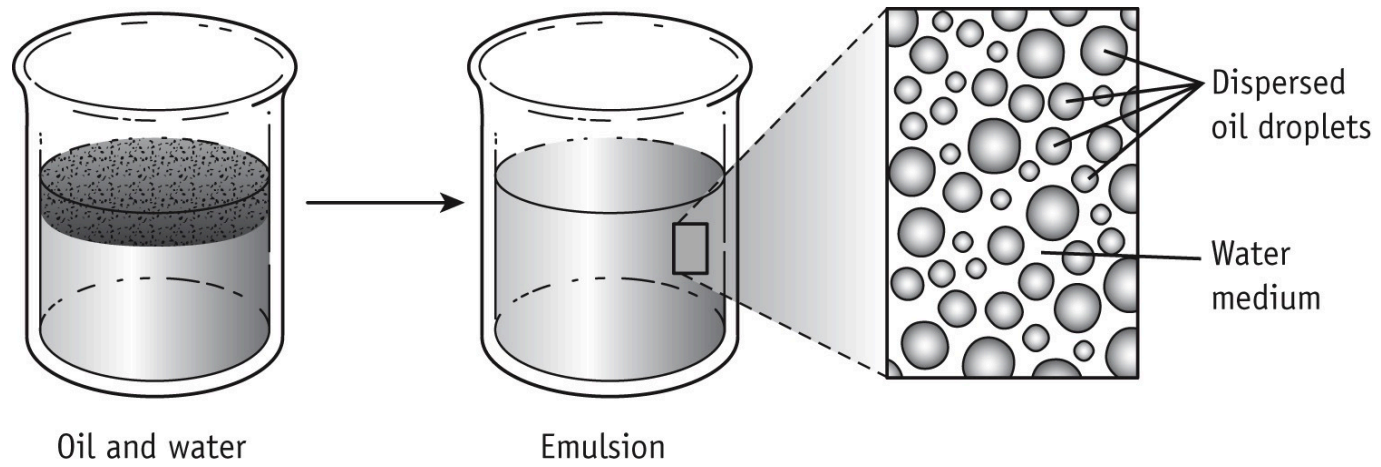
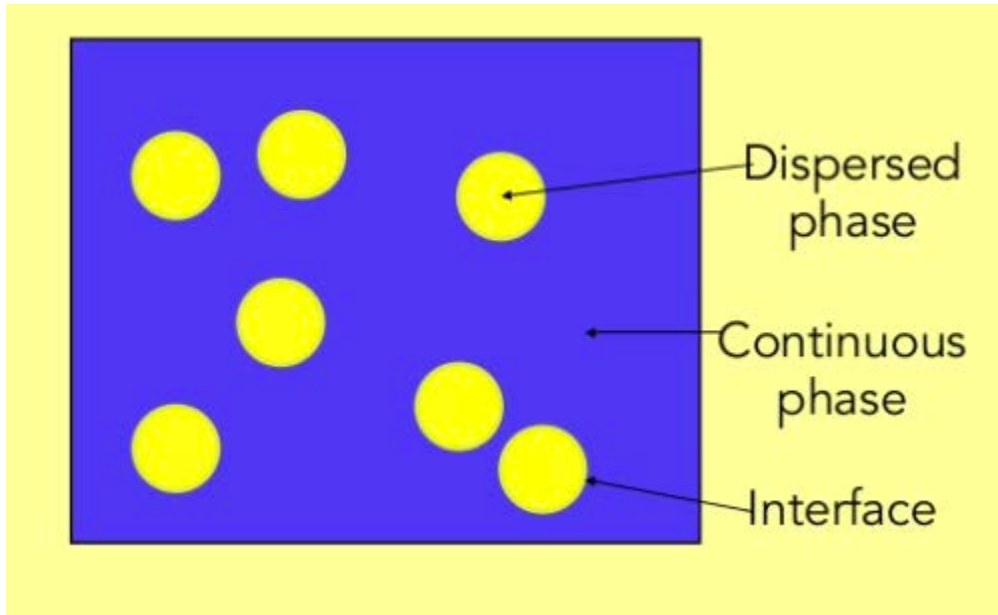


Figure 22-2 p455

Revisiting the technical terms

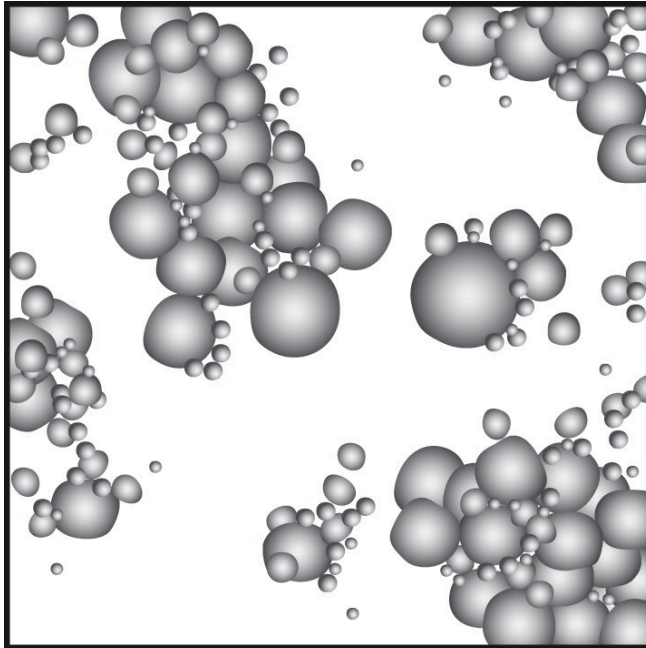


= Oil

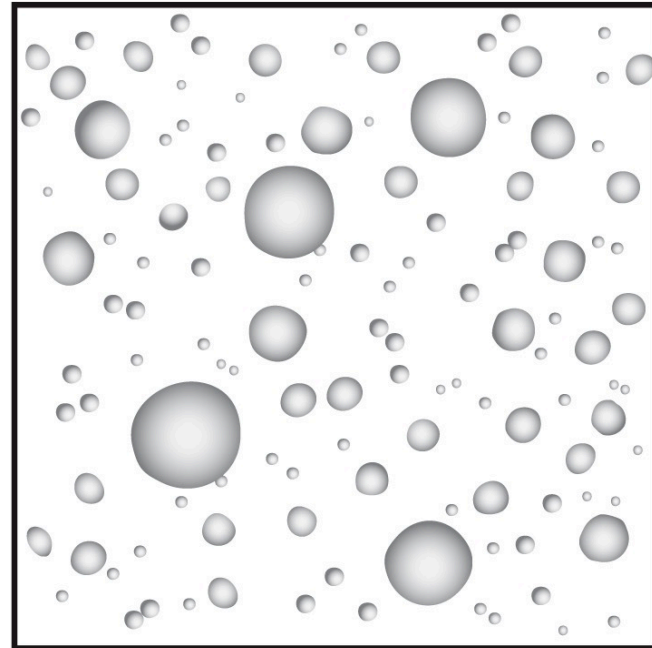
= Water

= Emulsifier

(Phospholipid from Egg)



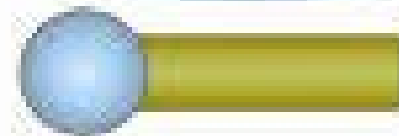
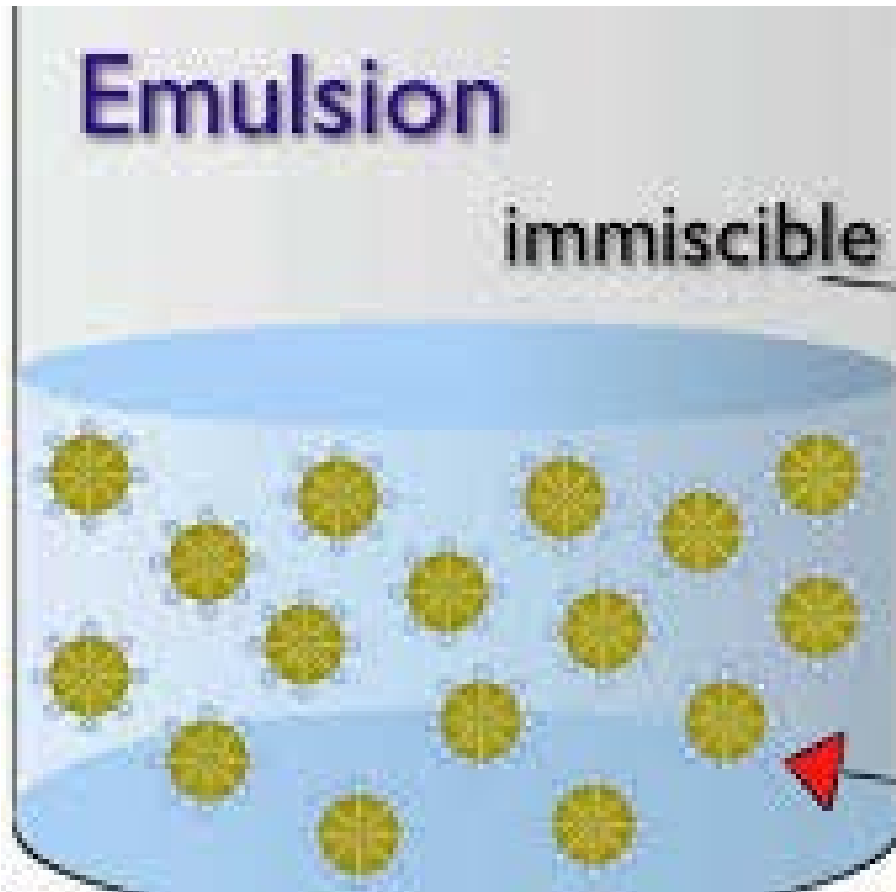
Without emulsifier



With emulsifier

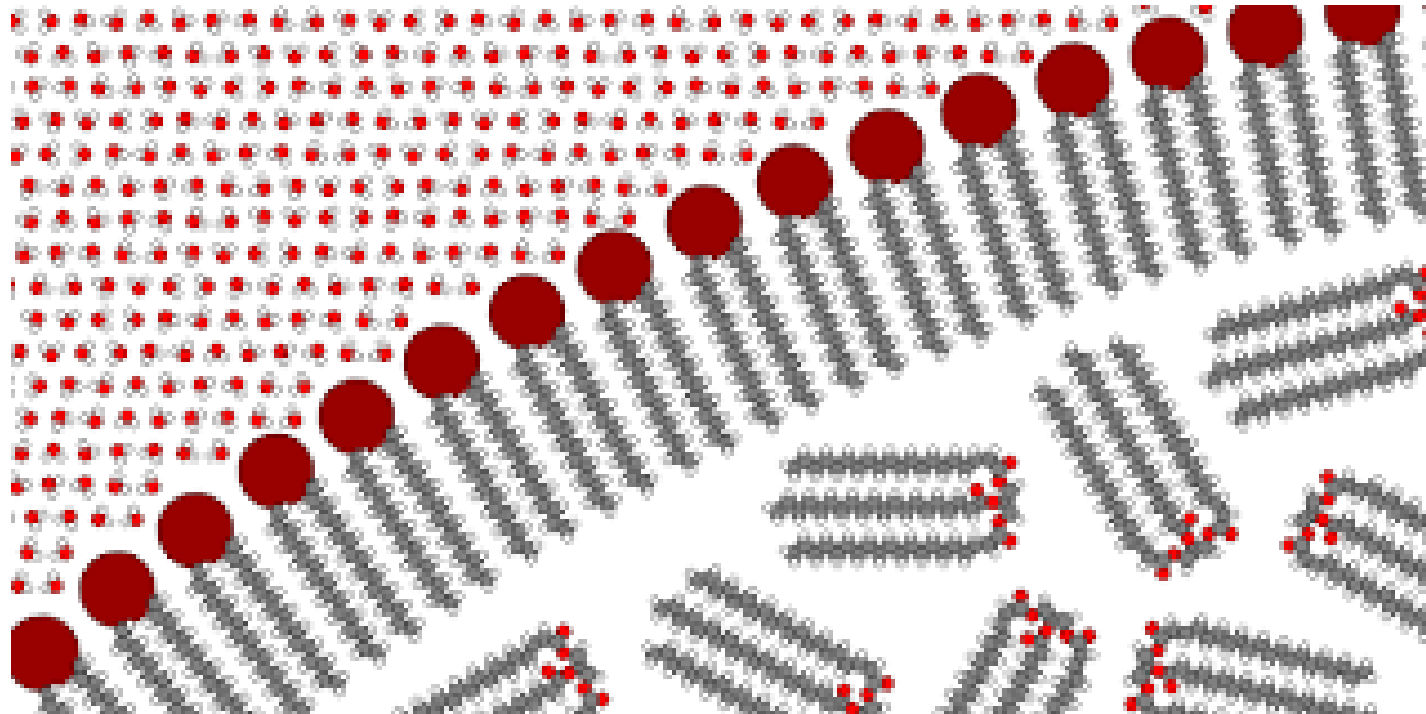
Emulsion

immiscible



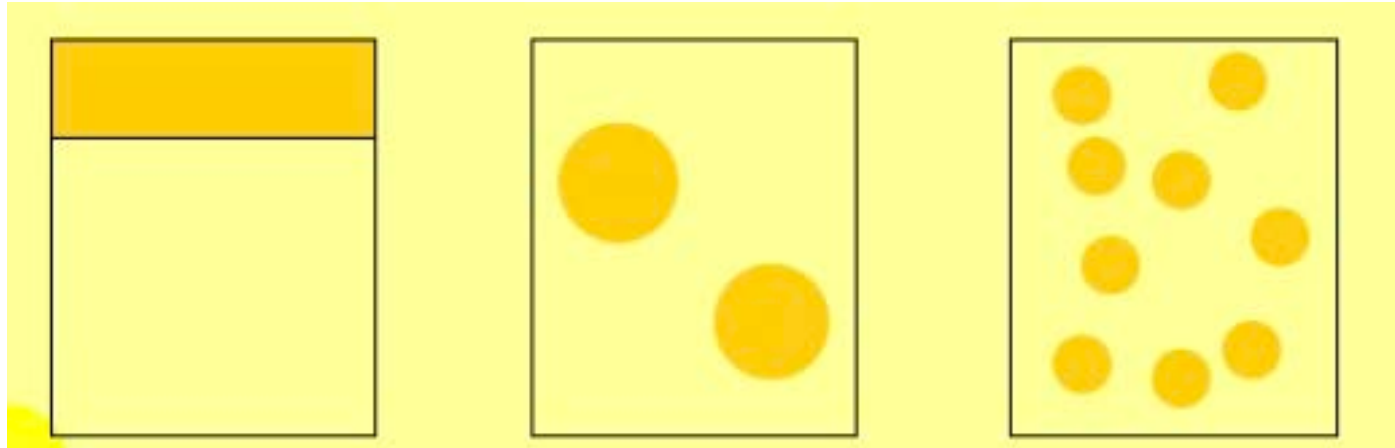
stabilised with
emulsifiers

Water & Hydrophilic Heads



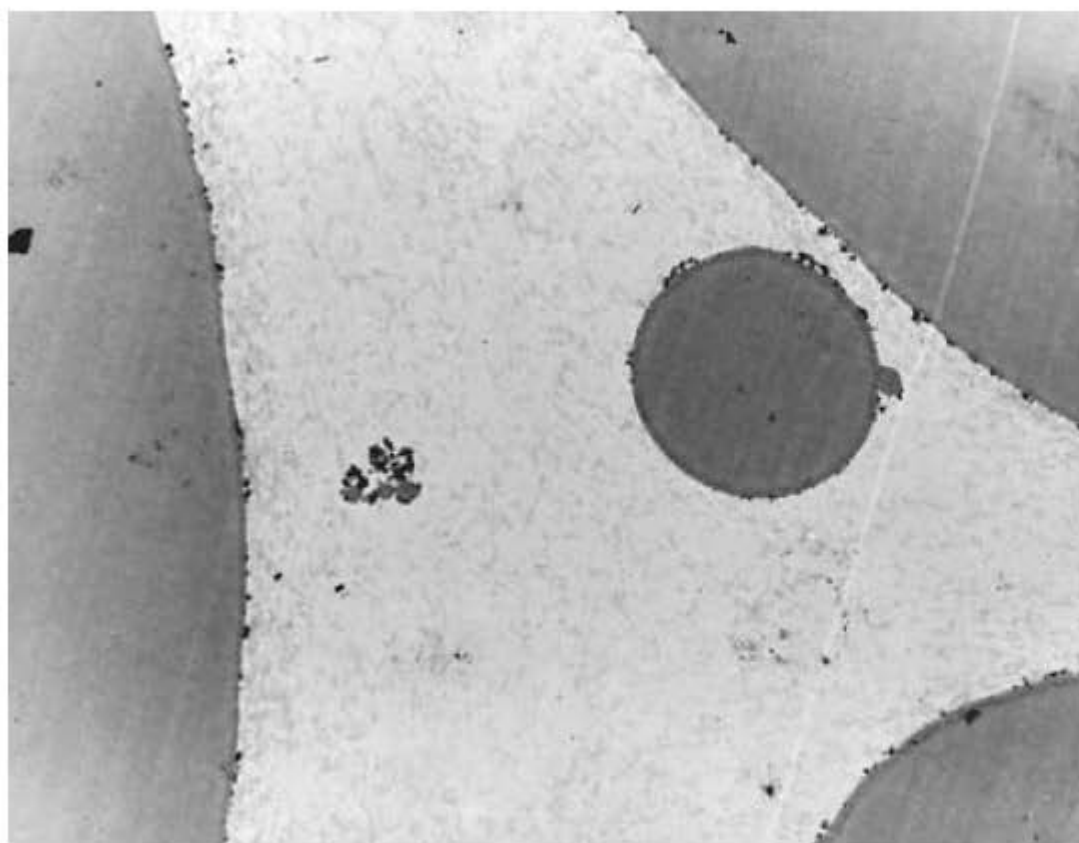
Lipids & Hydrophobic Tails

Emulsifiers help increase surface area contact



One category of emulsifiers are SURFACTANTS = Surface Active Agents

Mayonnaise: Electron Microscope View

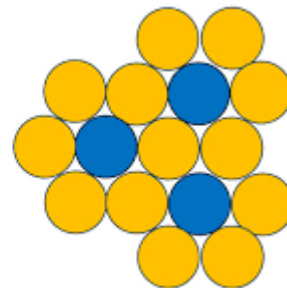


Oil droplets in mayonnaise. A view through an electron microscope. Protein and emulsifier molecules and aggregates, all from egg yolk, are present between the large droplets and on their surfaces, and help prevent them from coalescing.

It works both ways

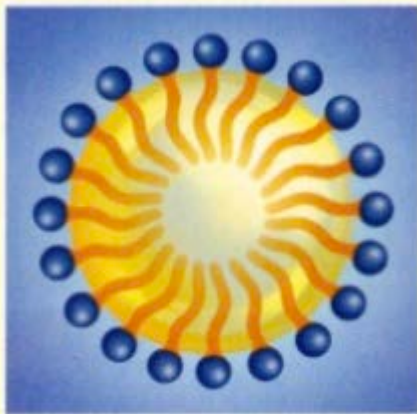


Oil-in-Water Emulsion

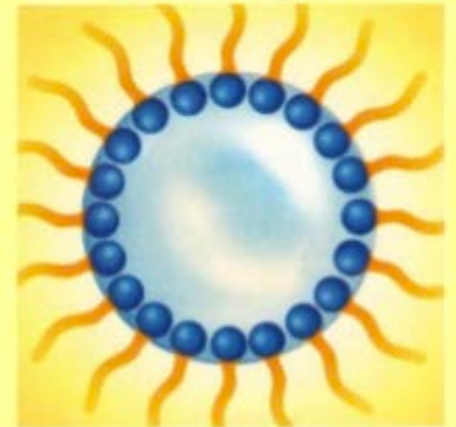


Water-in-Oil Emulsion

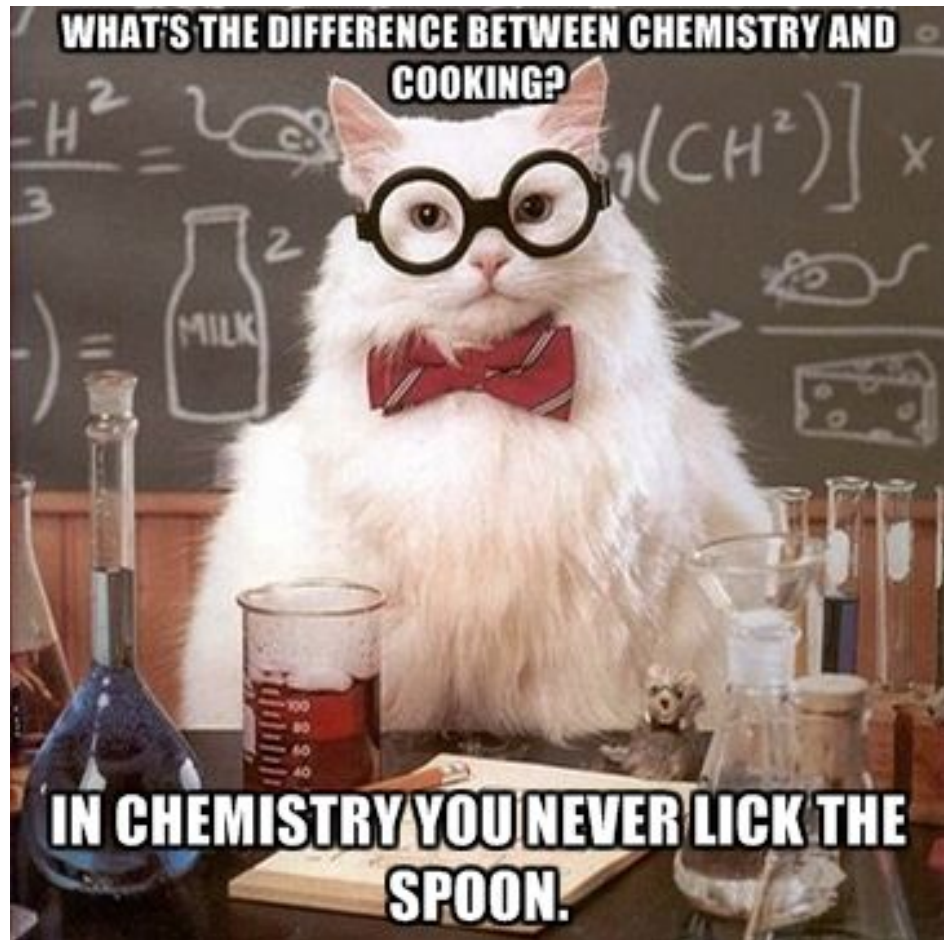
O/W emulsion



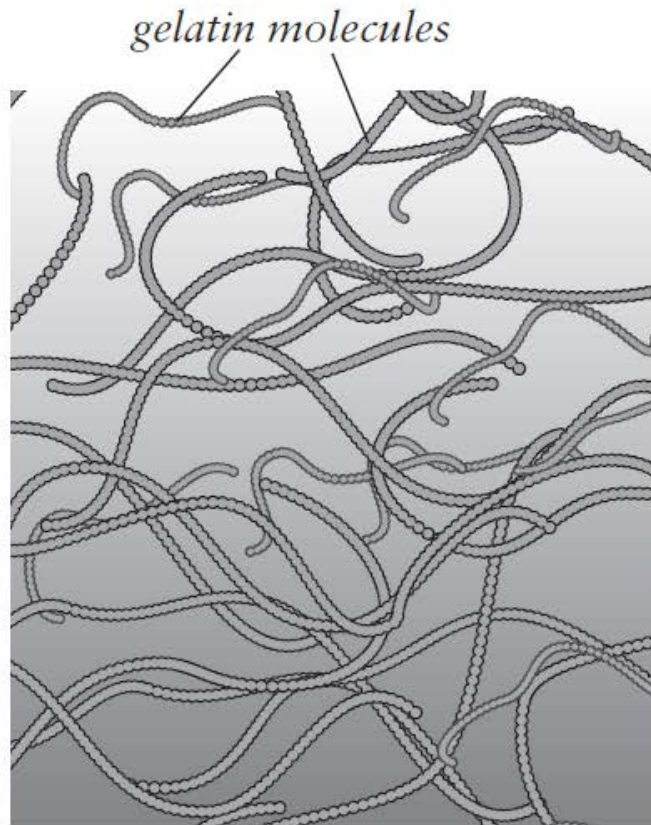
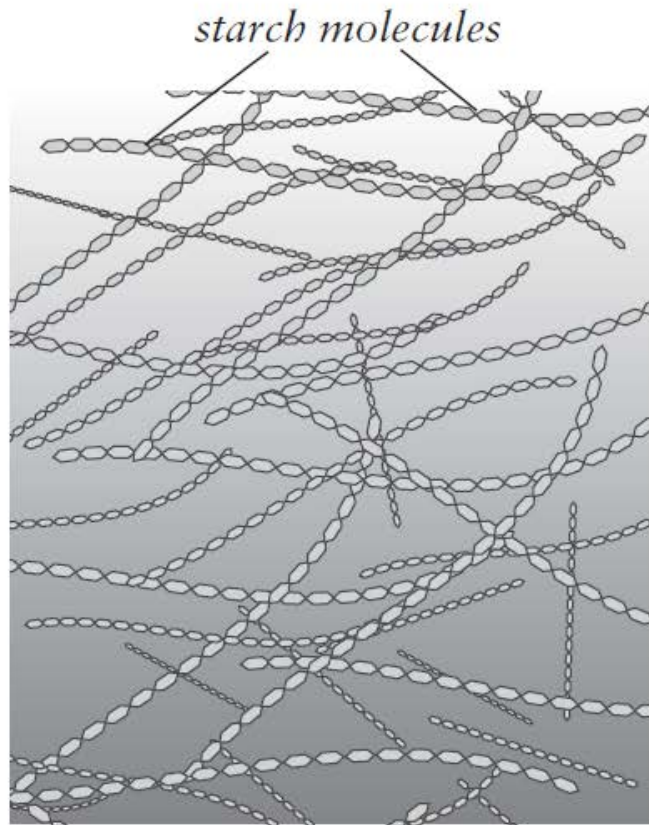
W/O emulsion



What's Cooking?

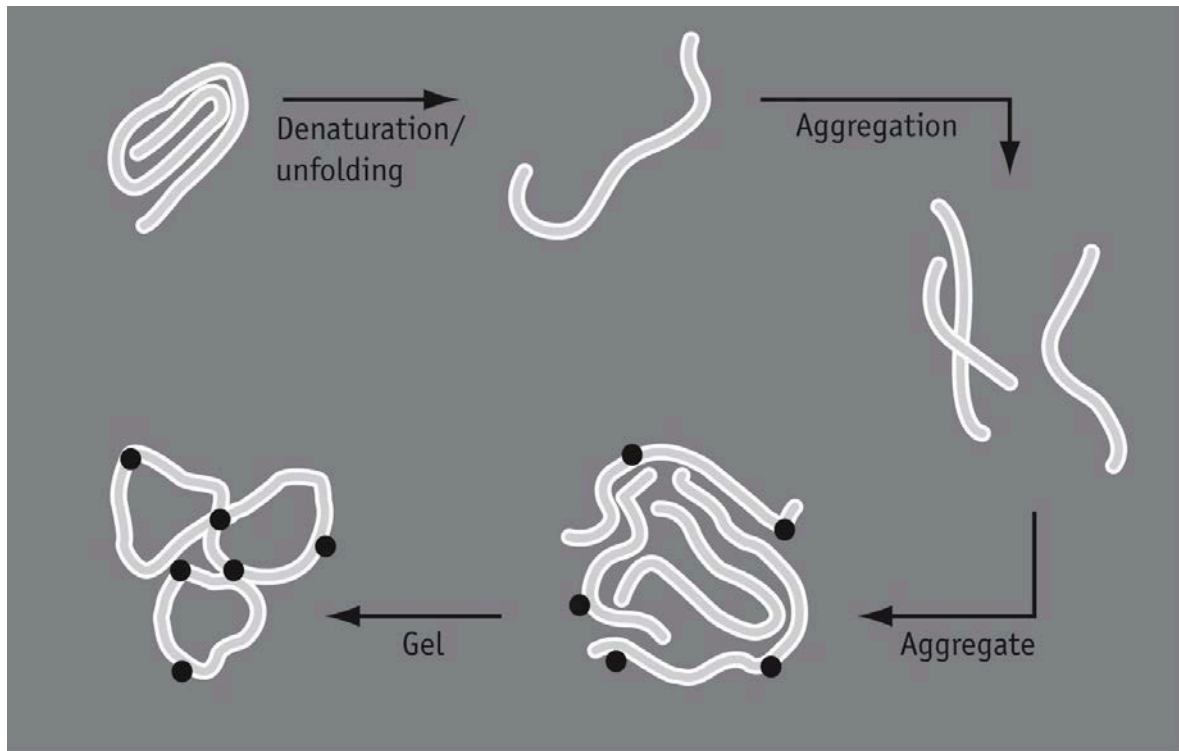


Thickening with long molecules: Gelatin and Starch

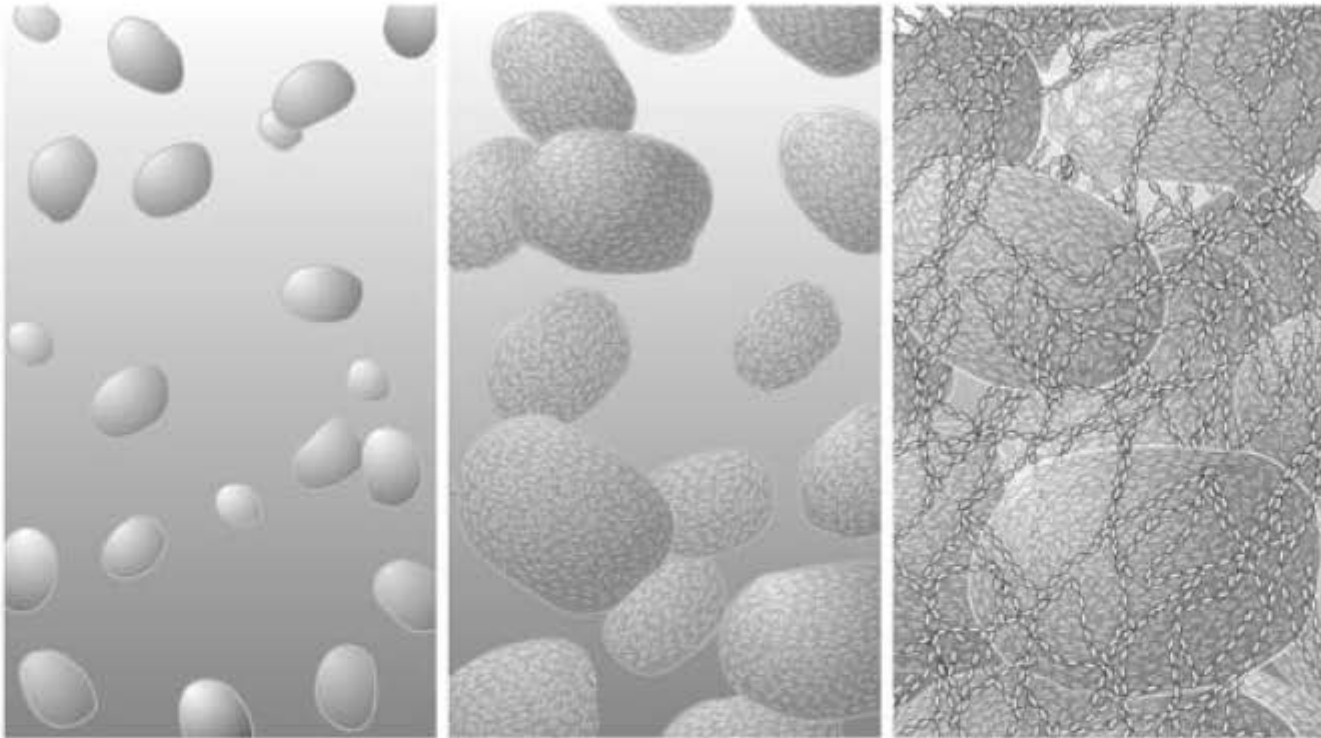


Thickening a liquid with long food molecules. Dissolved molecules of plant starch or animal gelatin get tangled up with each other and impede the flow of the liquid.

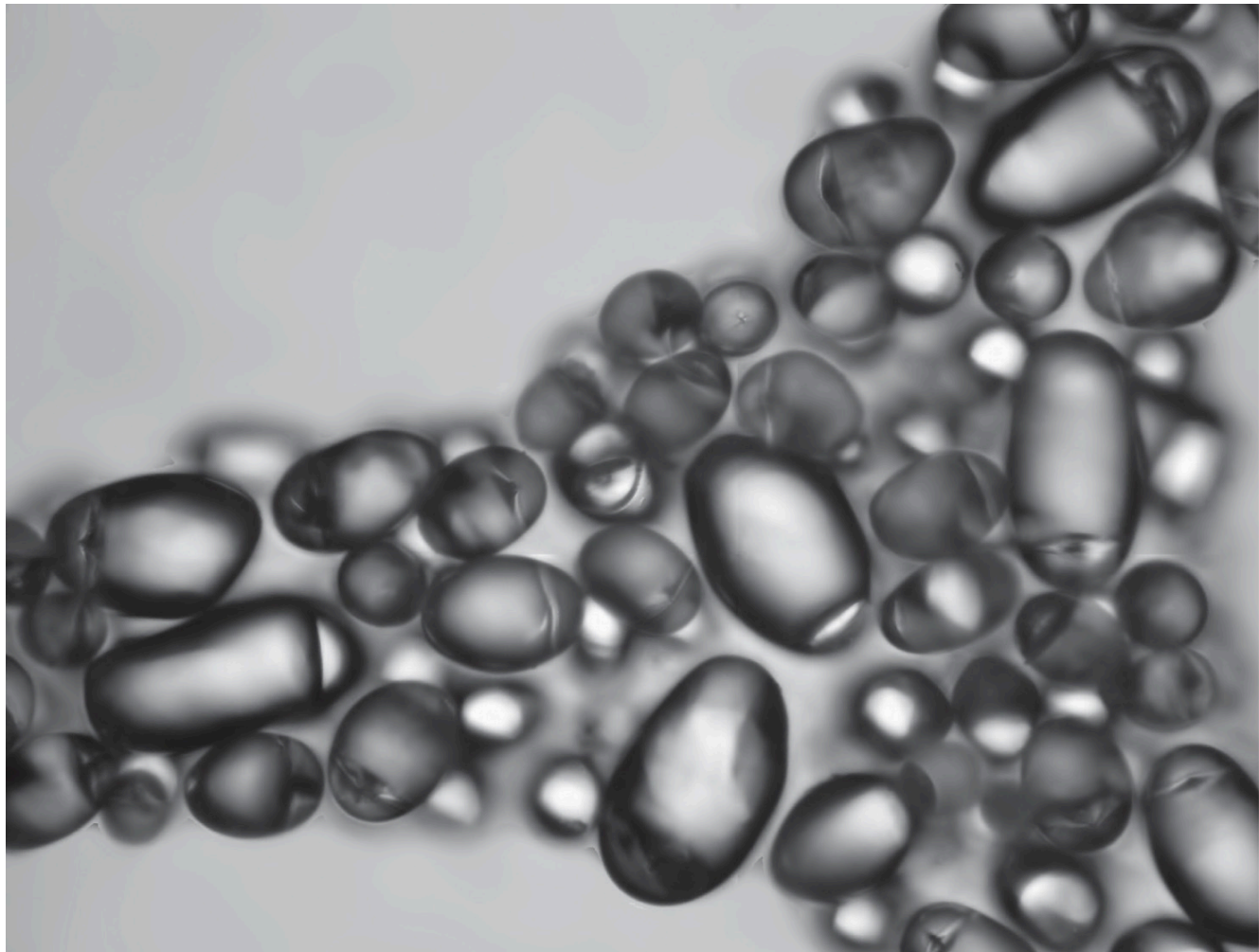
Gelatin



Starch

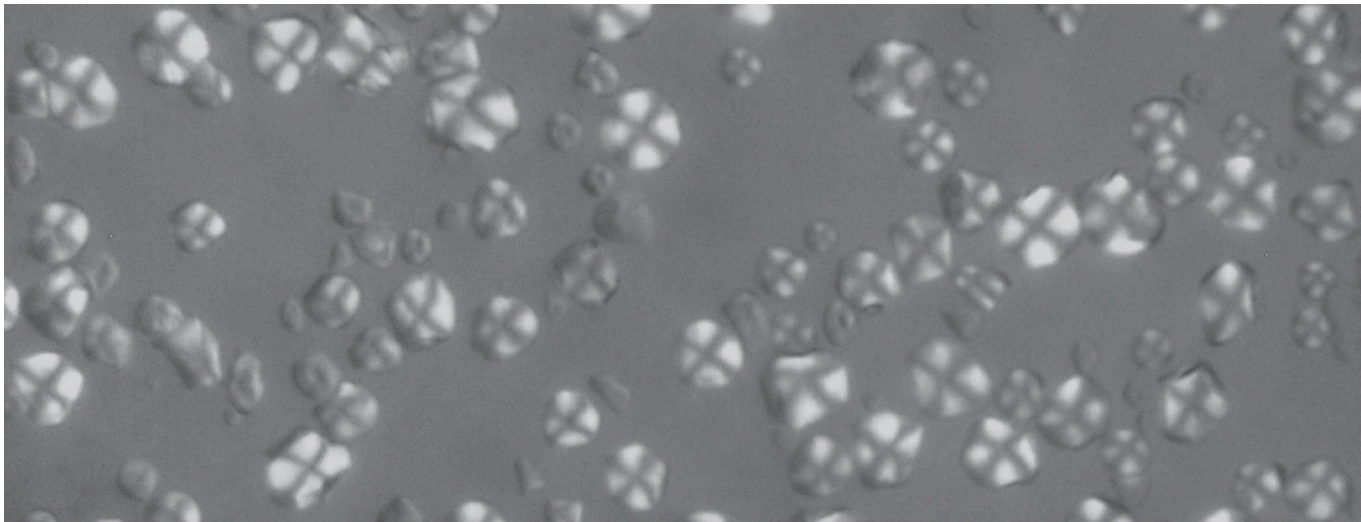


Thickening a sauce with starch. Uncooked starch granules offer little obstruction to the flow of the surrounding liquid (left). As the sauce heats up and the temperature reaches the gelation range, the granules absorb water and swell, and the sauce consistency begins to thicken (center). As cooking continues and the temperature approaches the boil, the granules swell even more and leak starch chains into the liquid (right). It's at this stage that the sauce reaches its maximum thickness.



Jubal Harshaw/www.shutterstock.com

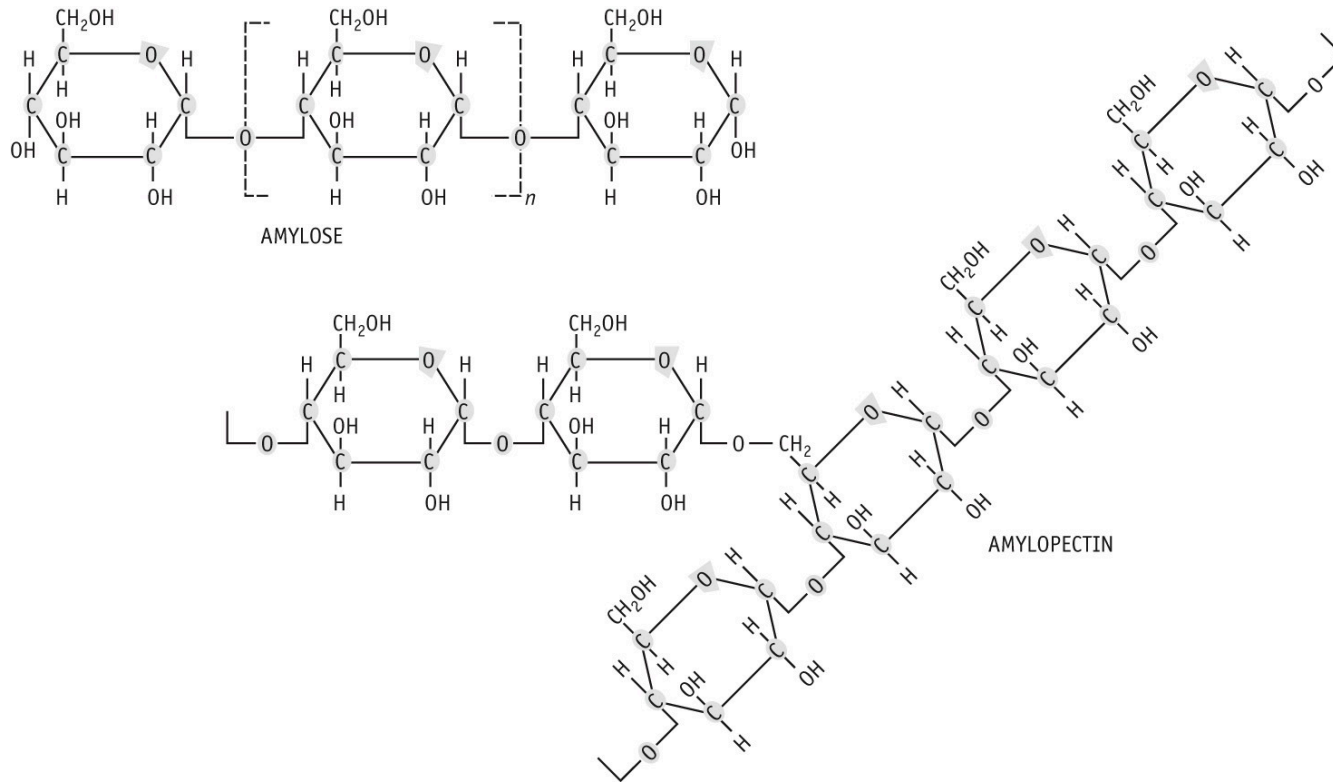
Figure 18-1 p390



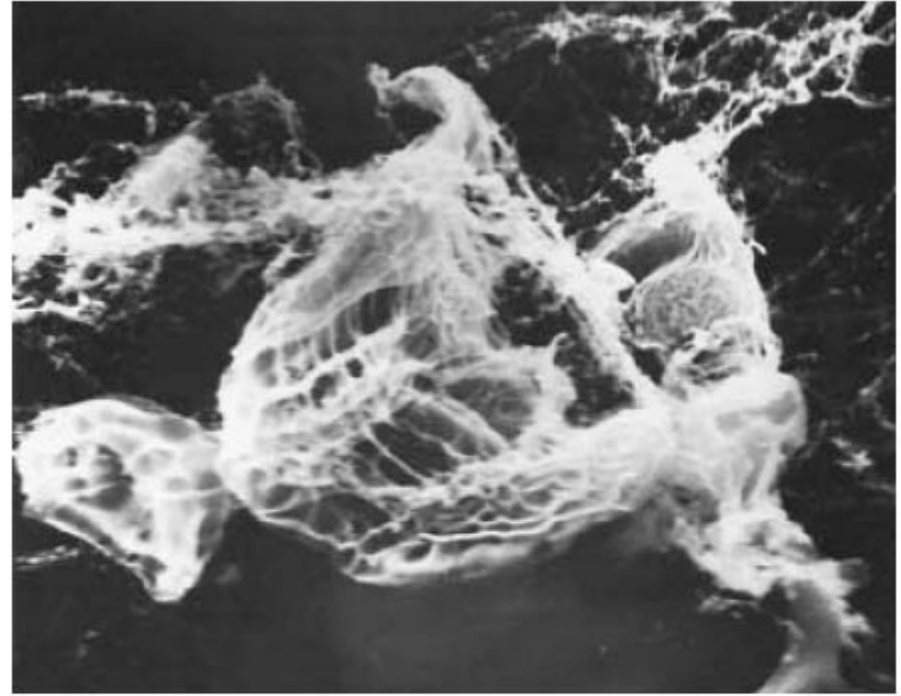
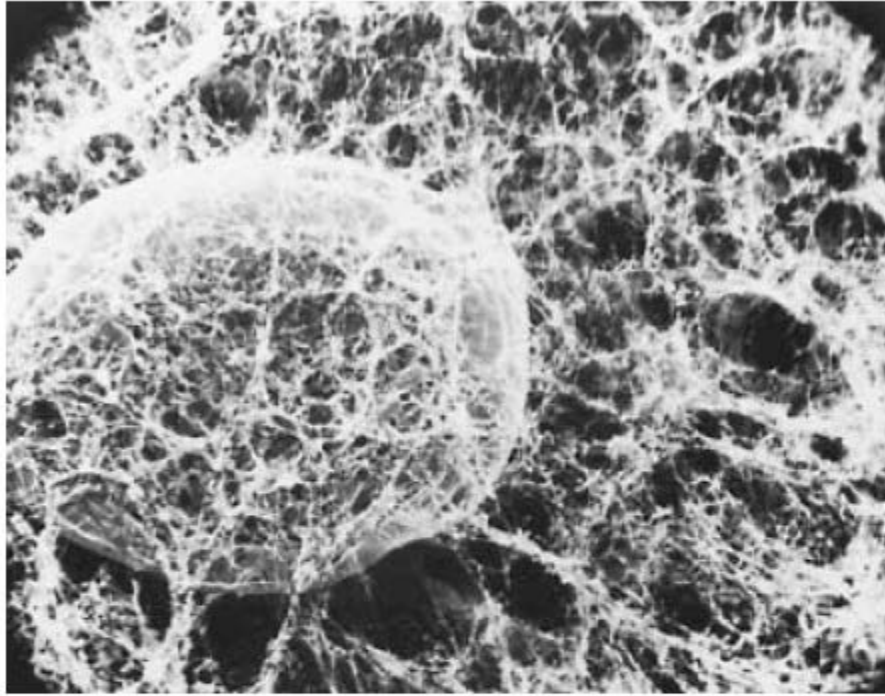
Source: Jan Homann

Figure 18-5 p394

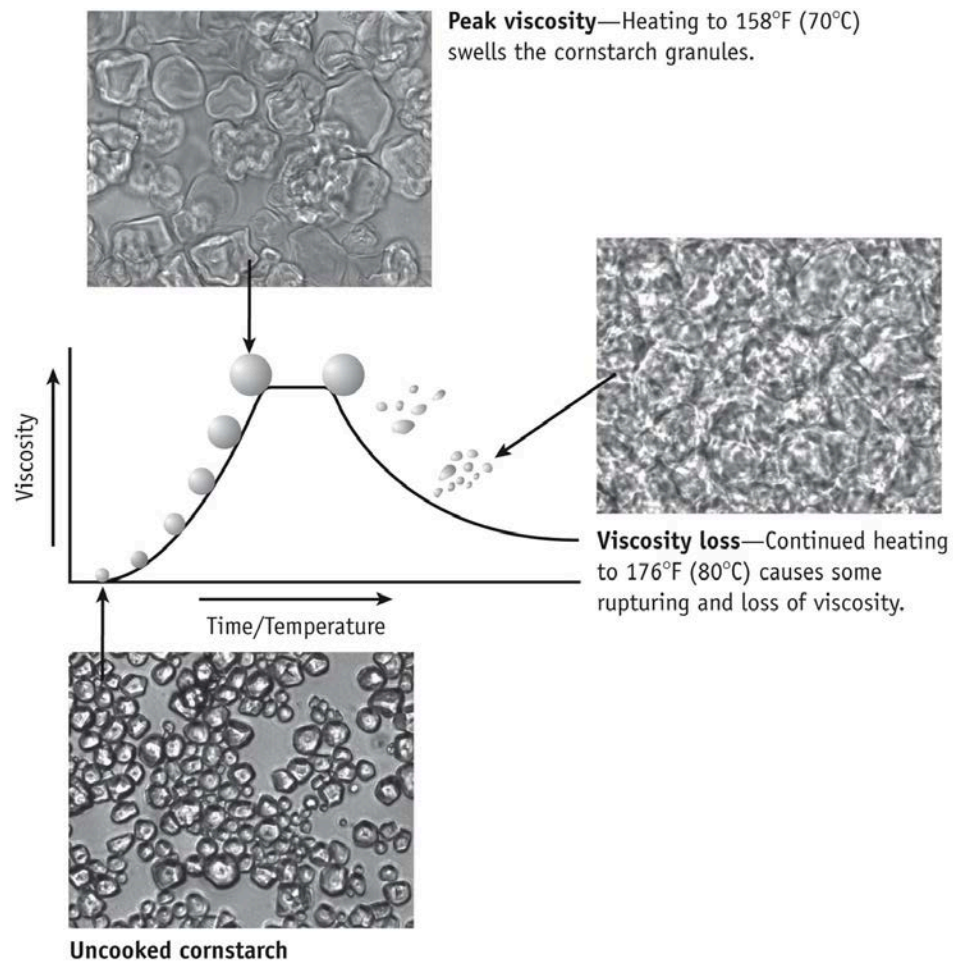
Two types of Starch



Starch: Swellin' & Gellin'



Starch in sauce making. A swollen granule of potato starch caught in a meshwork of molecules freed from it and other granules (left). A starch-thickened sauce is thickest at this stage, when both starch granules and molecules block the movement of water. A granule of wheat starch that has lost nearly all of its starch molecules to the surrounding liquid (right). As the granules in a starch-thickened sauce disintegrate, they no longer get caught in the mesh of free starch, and the sauce thins out.



Reprinted with permission from Ratnayake, Wajira S. and Jackson, David S., "Gelatinization and Solubility of Corn Starch during Heating in Excess Water: New Insights" Faculty Publications in Food Science and Technology. Paper 118. Copyright (2006). American Chemical Society.

Figure 18-4 p393



EGGS!

EGGS!

EGGS!

EGGS!

EGGS!

EGGS!

How many calories are the
judges about to consume?

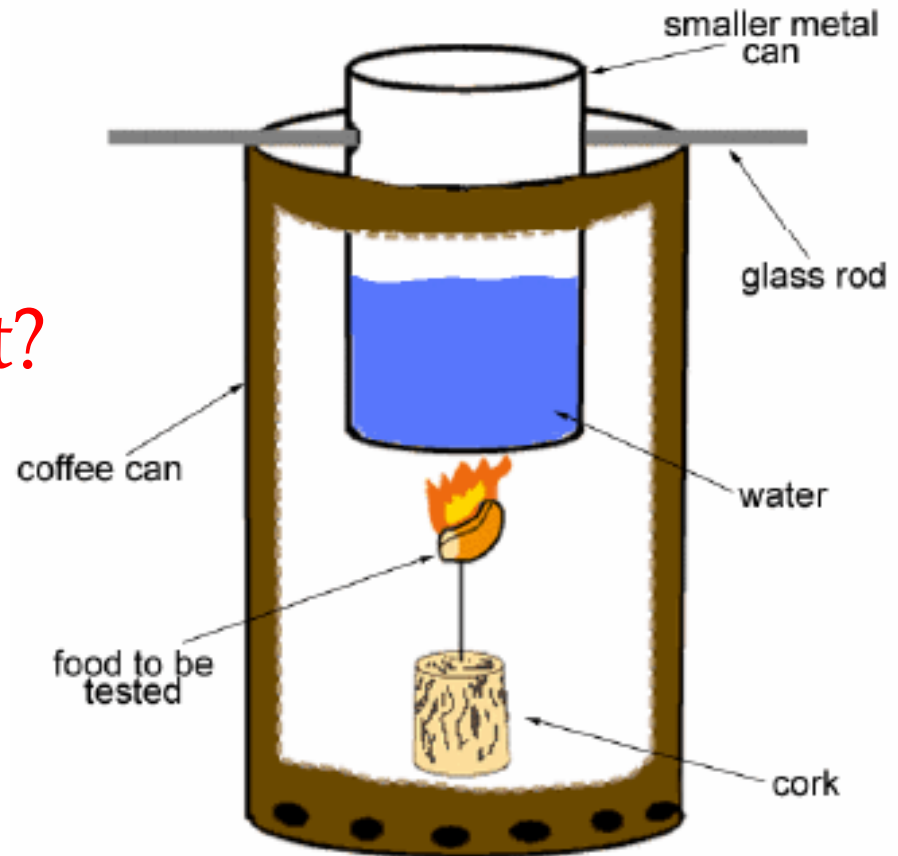
How many calories are the
judges about to consume?

What is a calorie?

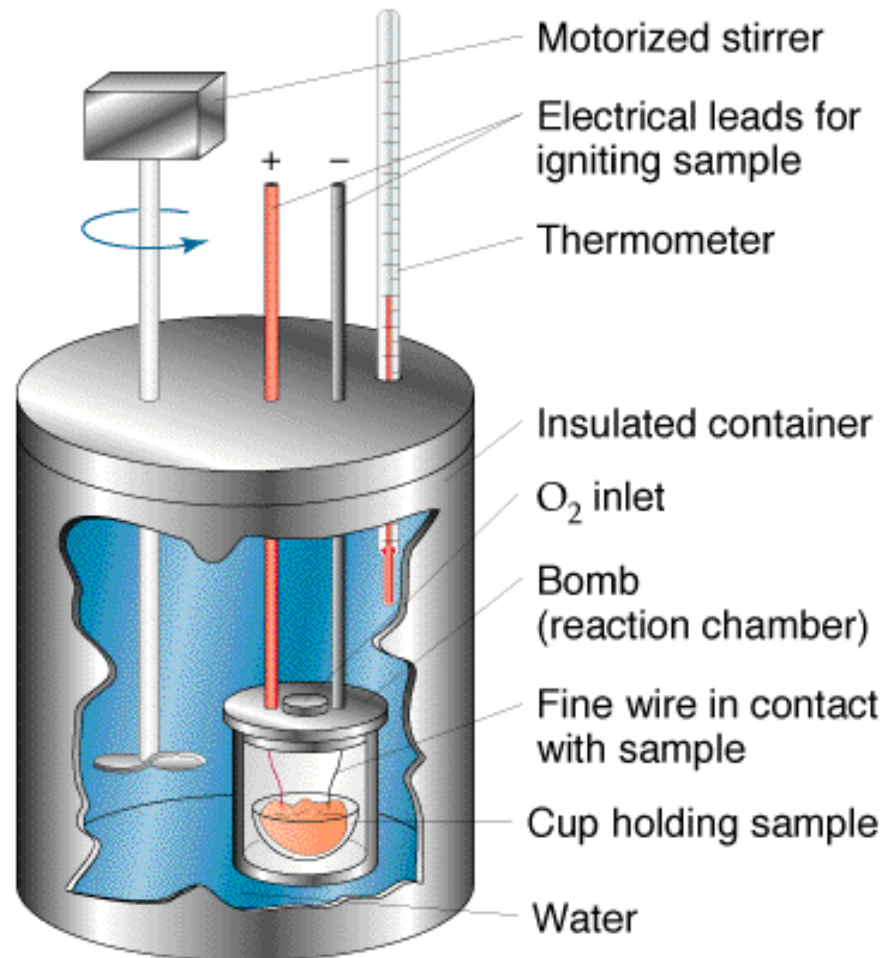
How many calories are the judges about to consume?

What is a calorie?

How do we measure it?



How is Heat Transfer Determined in Research Labs?



Bomb Calorimeter

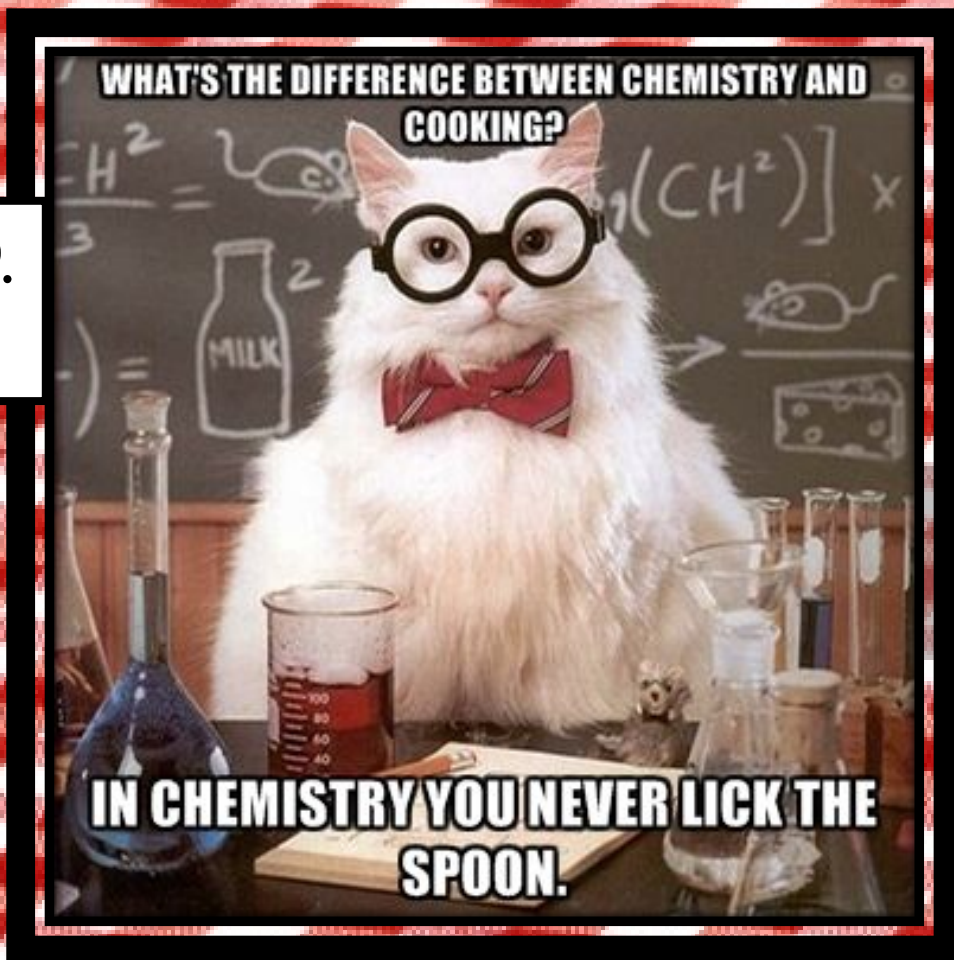
Judging Criteria

Criteria	Score
Taste (1 to 4)	
Originality (1 to 4)	
Use of secret ingredient (1 to 4)	
Scientific interest (1 to 4)	
Visual Presentation (1 to 4)	
Sub total	

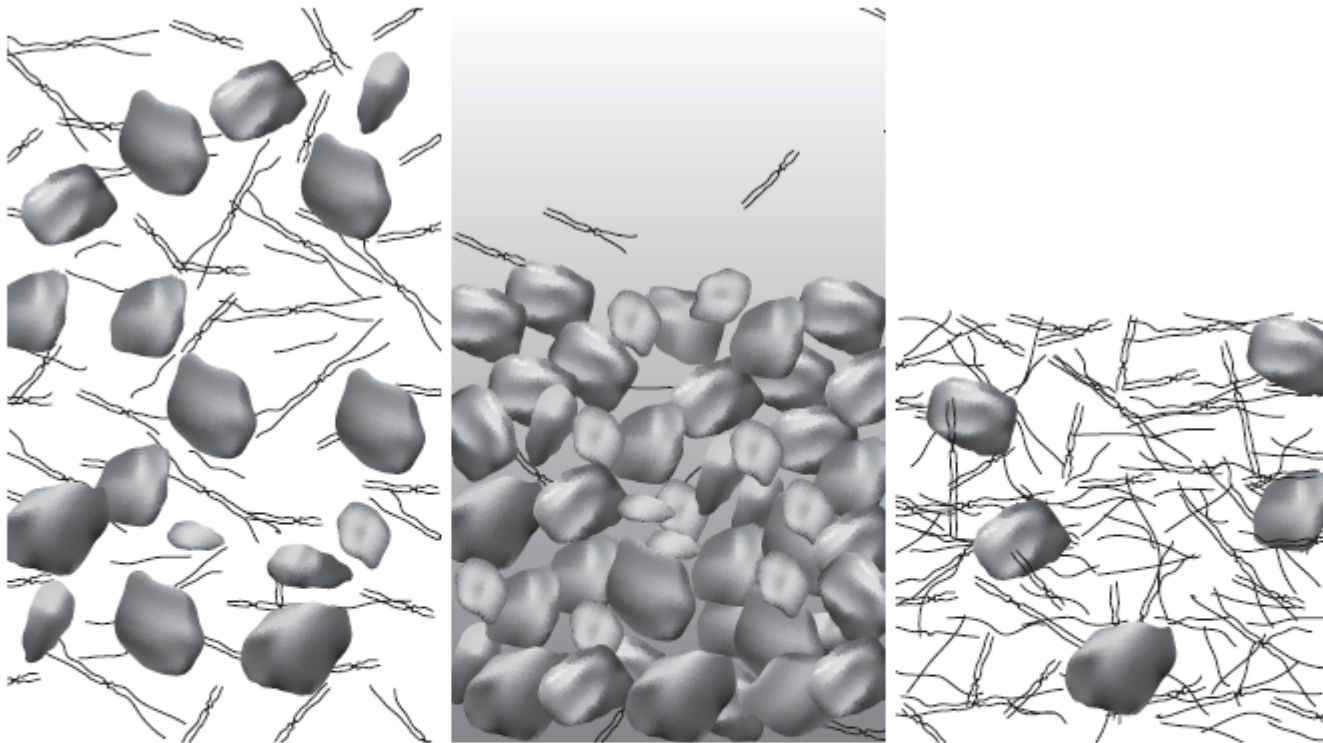
Culinary Chemistry: A Campus Cuisine Cookoff

MSUB Library Lecture series
Spring 16

Michele McMullen R.D.
Dr. Matt Queen



Puree: Smashing cells



A fruit or vegetable puree. Grinding plant tissue turns it inside out, freeing the cell fluids and breaking the cell walls and other structures into small particles. A puree is a mixture of plant particles and molecules floating in water (left). If left to stand, most purees will separate, with the larger particles settling to the bottom (center). This separation can be prevented, and the puree consistency thickened, by cooking the puree down and evaporating the excess water (right).



(1) Sauté aromatic vegetables (onions, garlic, shallots, or leeks) over medium heat. Adding herbs such as thyme, sage, or rosemary improves flavor.



(2) Pour in the liquid (water, broth, milk, or a combination) and heat to a simmer. Use 5 to 6 cups of liquid for every 2 pounds of main vegetable to be added.



(3) Add a thickener in the form of rice ($\frac{1}{2}$ cup) or raw potato (sliced). Starchy ingredients thicken when they gelatinize, making the soup feel creamy.



(4) The time to add the main vegetable (in even, bite-size pieces) is when the time remaining to cook the thickener is equal to how long it will take to cook the vegetable.



(5) When the thickener reaches its maximum viscosity, purée the soup in batches using a blender or food processor. Another option is to strain the soup to obtain a finer texture.



(6) Finishing touches include adding a little liquid to thin the purée if it is too thick, and/or adding a small amount of cream to give it a "cream" flavor. Serve hot or cold.

Digital Works

Figure 15-4 p332

TABLE 18-5 Selected Small Sauces Derived from Mother Sauces

Mother Sauces	Base Ingredient	Small Sauces
Béchamel (white) sauce	Milk	Cheddar cheese Cream Mornay Mustard
Espagnole (brown) sauce	Brown stock	Mushroom
Hollandaise sauce	Butter	Maltaise Mousseline
Tomato sauce	Tomato	Creole Portuguese Spanish
Velouté sauce	White stock Chicken stock Fish stock	Curry Mushroom Herb

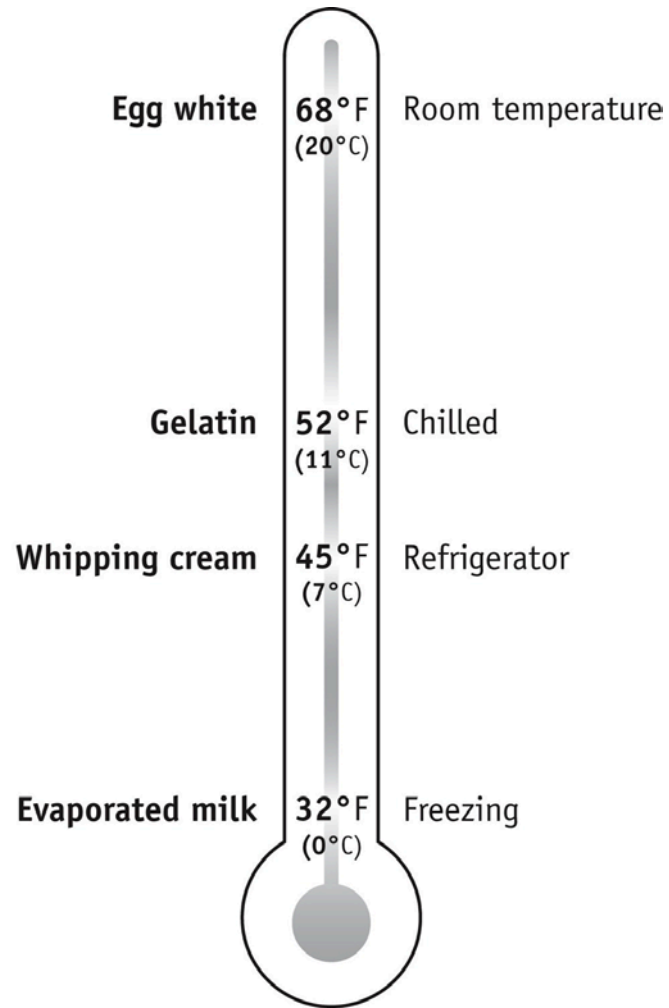
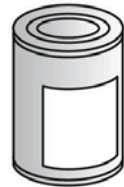
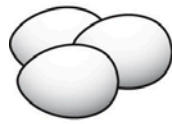


Figure 10-9 p227